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1	Pesticide biosensors: trends and progresses
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9 10	Keywords: Pesticides, biosensors, enzyme inhibition-based biosensor, immunosensors, aptasensors.
11	
12	Abstract
13	Pesticides, chemical substances extensively employed in agriculture to optimize crop yields,
14	pose potential risks to human and environmental health. Consequently, regulatory frameworks
15	are in place to restrict pesticide residue concentrations in water intended for human
16	consumption. These regulations are implemented to safeguard consumer safety and mitigate
17	any adverse effects on the environment and public health. Although Gas Chromatography- and
18	Liquid Chromatography- Mass Spectrometry (GC-MS and LC-MS) are highly efficient
19	techniques for pesticides quantification, their use is not suitable for real-time monitoring due to
20	the need for sophisticated laboratory pretreatment of samples prior to analysis. Since they would
21	enable analyte detection with selectivity and sensitivity without sample pretreatment,
22	biosensors appear as a promising alternative. These consist of a bioreceptor allowing for
23	specific recognition of the target and of a detection platform, which translates the biological
24	interaction into a measurable signal. As early detection systems remain urgently needed to

promptly alert and act in case of pollution, we review here the biosensors described in theliterature for pesticide detection to advance their development for use on field.

27 Introduction

Pesticides are essential for agriculture to increase crop yields and to provide suitable food 28 production levels. Since their launch, they enabled substantial economic benefits and are thus 29 widely utilized [1]. However, their widespread use concomitantly affects our health [2–5] and 30 our environment [6, 7]. The International Agency for Research on Cancer (IARC) plays a 31 crucial role in assessing the impact of pesticides on human health. As a renowned authority in 32 identifying carcinogenic agents, it provides valuable information to decision-making bodies 33 responsible for alerting and safeguarding the public from cancer-causing agents found in food, 34 the environment, and workplaces. Notably, the IARC has classified s-metolachlor, a pesticide 35 36 of particular concern in France, as probably carcinogenic. In response, the French Agency for Food, Environmental and Occupational Health & Safety (ANSES) has taken steps to phase out 37 38 its primary use in phytopharmaceuticals [8]. Additionally, the IARC has classified other pesticides, such as captafol (a fungicide used in seed treatments), chlordane, dichlorvos, and 39 heptachlorophenol, as probably carcinogenic [9]. These assessments highlight the importance 40 of rigorous monitoring and control measures to mitigate the potential health risks associated 41 with these substances. 42

The use of pesticides also leads to the contamination of surface waters [6, 7, 10]. Over the last 20-25 years [11], public concerns about the presence of pesticide residues in food and drinking water raised, leading to the implementation of strict regulations by legislative authorities in order to control the quality of consumer products and especially of drinking water. According to the 98/83/EC EU Directive, specific regulations have been implemented regarding the quality limits of pesticides active substances and their relevant metabolites in water. The general quality limit is set at 0.1 μ g L⁻¹ per individual substance for most pesticides. However, certain highly

toxic substances such as aldrin, dieldrin, heptachlor, and heptachlor epoxide are currently 50 banned, and their quality limit is set at a stricter level of 0.03 µg L⁻¹. Additionally, to address 51 the potential simultaneous presence of multiple pesticides and relevant metabolites, a 52 cumulative approach is adopted. The sum of concentrations of all pesticides and relevant 53 metabolites in water should not exceed 0.5 µg L⁻¹, considering the combined effect of these 54 substances. In parallel, the EU Directive No 396/2005 ensures the quality of food products by 55 setting Maximum Residue Limits (MRLs) for pesticides that may be present. These MRLs are 56 continuously updated to accommodate the introduction of new pesticides on the market, 57 reflecting the dynamic nature of agricultural practices. 58

59 These regulations are designed to ensure the safety and quality of water resources, considering60 both individual substances and their potential cumulative impact on water quality.

To detect pesticides at the concentration limits set by the competent authorities, sophisticated 61 62 techniques are required. Gas chromatography and liquid chromatography coupled with mass spectrometry (GC-MS and LC-MS) are classical techniques used for the sensitive, selective and 63 reproducible analysis of pesticide residues in water [12–15]. Although these techniques are well 64 suited for the quantification of pesticides, the samples studied must be pre-treated and then 65 analyzed in sophisticated laboratories, which renders these techniques unsuitable for survey 66 67 analyses of running water that require monitoring at a suitable frequency to detect an eventual pollution peak in real time and to promptly take an action. In order to meet these requirements, 68 increasing efforts are devoted to explore alternative measures to detect pesticides with 69 selectivity and sensitivity. Biosensors appear as elegant tools to replace conventional methods 70 as they could enable the detection of analytes in real time, minimizing tedious sample 71 pretreatments. Table 1 shows a detailed comparison of critical properties between classical 72 techniques and biosensors employed for the detection of pesticides. 73

74

75 **Table 1** Comparison of characteristic parameters between traditional analytical techniques and

76	biosensors emplo	yed for the	detection of	of pesticides.
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Parameters	Traditional Analytical Techniques	Biosensors
Analysis time	Collection of samples, storage, and	Limited sample preparation, rapid
Cost	Sophisticated laboratory detection, high-tech equipment, trained laboratory personnel	Cost-effective, portable and simple in-situ detection
Eco-friendly and reusability	More organic solvent consumption, Not reusable	No organic solvent used, reusable
Sensitivity and selectivity	Highly sensitive and selective	Sensitive and specific
Commercialization	/	Limited commercial applications

A biosensor is the combination of a bioreceptor and of a detection platform (Figure 1). The
bioreceptor is a biological molecule that allows the specific recognition of the target, whereas

79 detection platform translates the biologic interaction into a measurable, optical, magnetic or

80 electrochemical signal [16–18].



81

82 Schematic illustration of the different types of bioreceptors and transducers used in a biosensor.

Biosensors [19–21]. In parallel, and due to their ease of use, simple detection techniques, and satisfactory sensitivity, optical biosensors are also largely used in biosensors developed for the detection of pesticides [20, 22]. Different types of biomolecules have been employed in biosensors, mainly antibodies, enzymes and aptamers [23]. These biomolecules, due to their nature, undergo specific recognition with their target by different types of interaction, notably electrostatic interactions, hydrogen bonding, aromatic ring stacking and van der Waals interactions. The bioreceptor ensures the specificity of the analysis, whereas the sensitivity is ensured by the detection platform. Biosensors, as point-of-care devices, offer advantages such as rapid detection, user-friendliness, accuracy, portability, cost-effectiveness and easy on-site detection [24, 25].

This review covers the bioreceptors the most used and investigated for biosensors development for on-site monitoring of pesticides, namely antibodies, enzymes and aptamers. Aptamers, which can be considered as chemical antibodies, are synthetic biological molecules that have been widely investigated in the last decade to develop biosensors for detecting pesticides. In this context, an in-depth description of the use of aptasensors as tools for pesticide detection is also reported here.

100

I. Enzyme-based biosensors

Owing to their high selectivity for their substrates to which they bind and drive catalytic 101 reactions, enzymes are of high potential for biosensors development. They are commonly used 102 103 for the detection of organophosphate (OP) and carbamate pesticides that inhibit the activity of acetylcholinesterase (AChE), an essential enzyme for the functioning of the central nervous 104 system [26]. The inhibition of AChE by OP and carbamate pesticides takes place via a 105 106 phosphorylation mechanism, which blocks serine in the active site through a nucleophilic attack and production of a serine phosphoester [27]. On the basis of this reaction scheme, the 107 electrochemical and optical recording of the enzymatic activity reduction enables the detection 108 of this particular type of pesticide in the matrices to be analyzed. Butyrylcholinesterase (BChE) 109 [28], tyrosinase [29, 30], alkaline phosphatase [31, 32], peroxidase [33], acid phosphatase [34] 110 urease [35] and exonuclease I [36] were also used in inhibition-based pesticide biosensors. As 111

described below, several enzyme-based biosensors were reported for pesticides detection, basedon electrochemical and optical transduction [37–39].

I.1. Enzymatic inhibition-based electrochemical biosensors 114 Electrochemical detection methods, including cyclic voltammetry, alternating current 115 116 voltammetry (ACV), electrochemical impedance spectroscopy (EIS), square wave voltammetry (SWV), differential pulse voltammetry, photoelectrochemical, and electrochemiluminescence, 117 enable the rapid and sensitive detection of redox-active target analytes without requiring 118 complex sample pre-treatment [40]. These electrochemical techniques generate target-induced 119 signals specifically triggered by the presence of the analyte, allowing for its accurate detection 120 and quantification. In particular, enzyme inhibition-based electrochemical biosensors have 121 122 emerged as a promising approach for pesticide detection (Figure 2(A)), enabling real-time and on-site monitoring of pesticide residues. These biosensors utilize the inhibition of enzymatic 123 activity to achieve precise and frequency-appropriate detection of pesticides, making them 124 valuable tools in pesticide analysis and environmental monitoring. 125





126

127 Comparative schematic diagram of AChE inhibition-based: (A) electrochemical, (B)
128 fluorometric, and (C) colorimetric biosensors for detection of organophosphorus compounds.
129 (B) Reprinted from [41], with permission from Elsevier. (C) Reprinted from [42], with
130 permission from American Chemical Society.

131 I.1.1. Differential pulse voltammetry

Chauhan et al. [37] have selectively detected several organophosphorus pesticides (malathion,
chlorpyrifos, monocrotophos and endosulfan) with a high sensitivity by covalently
immobilizing AChE onto iron oxide nanoparticles (Fe₃O₄NPs) and carboxylated multiwalled

carbon nanotubes (c-MWCNTs) modified Au electrodes. The acetylcholine (ATCI) is
enzymatically hydrolyzed by AChE into thiocholine that undergoes electrocatalytic oxidative
dimerization at +0.4 V vs. an Ag/AgCl reference electrode and produces a disulphide
compound. The oxidation of thiocholine at the working electrode surface (equations 1 and 2) is
correlated to the activity of AChE.

140 ATCI +
$$H_2O \rightarrow$$
 Thiocholine + CH_3COOH (1)

141 2 Thiocholine
$$\rightarrow$$
 S(CH₂)₂N⁺(CH₃)₃ - S(CH₂)₂N⁺(CH₃)₃ + 2H⁺ + 2e⁻ (2)

In the presence of pesticides, AChE activity is reduced involving a decrease of thiocholine production, resulting in a decreased electrical signal which is proportional to pesticides concentration. Under optimal conditions, the degree of inhibition caused by those pesticides was found to be directly proportional to their concentrations, within the following ranges: 0.1-40 nM for malathion, 0.1-50 nM for chlorpyrifos, 1-50 nM for monocrotophos, and 10-100 nM for endosulfan. The detection limits were determined to be 0.1 nM for malathion and chlorpyrifos, 1 nM for monocrotophos, and 10 nM for endosulfan.

Ma et al. [43] have described the fabrication of PtPd@NCS core-shell structured 149 nanocomposites consisting of a bimetal core (Pt and Pd) encased in an N-doped carbon shell 150 151 (NCS). The nanocomposites were prepared using a simple one-pot approach, involving reduction of metal salt precursors, self-polymerization of dopamine, and co-assembly of 152 Pluronic F127. The nanocomposites were then used to prepare an AChE-inhibition based 153 biosensor for detecting the following organophosphate pesticides: malathion, chlorpyrifos and 154 parathion methyl. The inhibition rate is defined following equation 3 where I₀ represents the 155 original signal recorded by DPV measurements with the as-prepared 156 AChE/PtPd@NCS/graphene carbon electrode (GCE) biosensor in phosphate buffer saline 157 (PBS, pH 7.5) containing ATCI (2.0 mM). I₁ represents the residual signal. 158

159 Inhibition (%) =
$$\frac{I_0 - I_1}{I_0} \times 100\%$$

(3)

The biosensor functions in a linear detection range for each pesticide, at extremely low limits 160 of detection in the range of femtomoles to attomoles (7.9×10^{-15} M, 7.1×10^{-14} M, and 8.6×10^{-15} M, 7.1×10^{-15} 161 10⁻¹⁵ M for malathion, chlorpyrifos, and parathion-methyl, respectively). These results 162 underscore the exceptional sensitivity of the AChE biosensor prepared using PtPd@NCS 163 164 nanocomposites. Recently, nanozyme-based biosensors also show tremendous potential for pesticide detection [44]. In their study, Wu et al. [45] presented an innovative electrochemical biosensor for 165 detecting organophosphate pesticides using a two-dimensional MnO₂ nanozyme by DPV measurements. 166 The researchers used manganese dioxide nanosheets (MnNS), more specifically two-dimensional (2D) 167 MnO₂ sheets, in conjunction with acetylcholinesterase (AChE) to create a homogeneous electrochemical 168 biosensor. This novel biosensor demonstrated excellent performance, exhibiting a linear response for 169 paraoxon in the 0.1 to 20 μ g L⁻¹ range, with a LOD of 0.025 μ g L⁻¹. Importantly, these results align with 170 171 the maximum residue limit (MRL) requirements established by the European Union (EU).

172

I.1.2. Potentiometry

Vaghela et al. [35] have described a potentiometric electrochemical biosensor based on bionanoconjugate of urease with gold nanoparticles (AuNPs) entrapped in agarose-guar gum to detect glyphosate. The action of urease catalyzes the formation of ammonium ions (equation 4) that are measured by their selective electrode. In the presence of glyphosate, the urease activity is inhibited, leading in a decrease of ammonium ions formation. According to this inhibition mechanism, the decrease in potentiometric signal indicates the presence of glyphosate and can be used for its detection.

$$H_2N \xrightarrow{O} H_2O \xrightarrow{Urease} H_2N \xrightarrow{-COOH} H_3$$
(4)

$$\underbrace{H_2O}_{180} \longrightarrow H_2CO_3 + 2 \operatorname{NH}_3 \longrightarrow \operatorname{HCO}_3 + \operatorname{NH}_3 + \operatorname{NH}_4^+$$

181 According to this mechanism of detection, a linear response with respect to glyphosate 182 concentration is obtained in the range of 0.5 to 50 μ g L⁻¹, which is within the maximum residual 183 limit established by the World Health Organization (WHO).

184

I.1.3. Cyclic voltammetry

Five pesticides were detected by glutathione-S-transferase (GST) based electrochemical 185 biosensing: carbendazim, chlorpyrifos, dichlorodiphenyltrichloroethane (DDT), dinocap and 186 ethion [46]. Cyclic voltammetry was used to detect all five pesticides. GST was immobilized 187 onto platinum electrodes using a graphene oxide-gelatin matrix. The immobilization of GST 188 189 was confirmed through cyclic voltammetry, electrochemical impedance spectroscopy, scanning electron microscopy and chronoamperometry. The developed GST biosensor has several 190 advantages over other biosensors: it is cost effective as it is reusable for 8-10 consecutive 191 measurements, monitoring 192 and real-time demonstrated when using the was chronoamperometric mode. 193

194

I.1.4. Chronoamperometry

BChE, alkaline phosphatase and tyrosinase were used to develop a novel three-dimensional origami paper-based device that utilizes enzyme-inhibition to detect several pesticides: paraxon, 2,4-dichlorophenoxyacetic acid and atrazine [28]. This device combines two office paper-based screen-printed electrodes with multiple filter paper-based pads for loading enzymes and their substrates. By folding and unfolding the filter paper-based structure, the device can analyze pesticides without any need for reagents or sample treatment (**Figure 3**).



201

Schematic illustration of configuration and measurement procedure of the paper-basedplatform. Reprinted from [28], with permission from Elsevier.

The paper-based platform is advantageous due to its low cost, portability, and ability to analyze 204 multiple pesticides. However, it is based on enzyme inhibition and may not be suitable for 205 detecting certain types of pesticides. Zhao et al. [47] reported on an electrochemical sensor 206 based on AChE inhibition to detect paraxon. The mentioned biosensor is composed of a screen-207 208 printed carbon electrode (SPCE) modified with electrodeposited gold nanoparticles (AuNPs) on which MoS₂ nanosheets were immobilized. AChE was immobilized in the last step onto 209 210 MoS₂ via a glutaraldehyde crosslinker. Using amperometry under optimized conditions, paraxon could be detected at a very low limit of detection (LOD = $0.013 \ \mu g \ L^{-1}$). 211

I.2. Optical biosensors

Although enzyme-based biosensors relying on electrochemical detection are highly sensitive and could be used for real time detection of pesticides, the immobilization of enzymes on the electrode surface is generally tedious and time consuming. In order to overcome these difficulties, optical biosensors have been used as an alternative since they are reliable, easy to use, fast and reveal high sensitivity.

218

I.2.1. Fluorescence spectroscopy

219 A fluorometric AChE-based biosensor allowed the rapid, simple and sensitive detection of parathion-methyl (Figure 2(B)) [41]. Gold nanoparticles immobilized on the surface of 220 NaYF4:Yb,Er up-conversion nanoparticles (UCNPs) enable fluorescence resonance energy 221 222 transfer (FRET). Thiocholine present in the solution, after hydrolysis of acetylthiocholine by AChE, interacts with the AuNPs via electrostatic interactions preventing the formation of 223 AuNPs-UCNP complexes, which leads to an increased fluorescence signal. In the presence of 224 parathion-methyl, AChE activity is inhibited, thus preventing the production of thiocholine, 225 which favors the FRET phenomenon, and thus the decrease of fluorescence. The biosensor 226 developed in this study exhibited high sensitivity and stability, indicating its potential for 227 detecting organophosphate pesticides in real samples. Moreover, the biosensor preparation 228 process is less time-consuming compared to grafted enzyme biosensors as the enzyme is not 229 immobilized onto the nanoparticle surfaces but only adsorbed. This advantageous feature 230 makes the biosensor a promising tool for practical applications in detecting pesticides. 231

232

I.2.2. Colorimetric assay

The rapid and onsite detection of several pesticides remains challenging, like glyphosate, which is of high polarity, has metal-chelating properties and interfere with organic substances in the environment. Besides, its similarity with its by-products renders its detection difficult. Enzymebased methods along with colorimetric detection reveal several advantages matching with glyphosate analysis, i. e. simple preparation, rapid detection and ease of results acquisition [48, 49]. Luo et al. [42] have reported a novel colorimetric nanozyme sheet for the rapid detection
of glyphosate (Figure 2(C)). Physically adsorbed peroxidase enzymes catalyze the oxidation
of chromogenic substrates to produce a color change, the intensity of which is measured by
UV-Visible spectrophotometry. In the presence of glyphosate, peroxidase is inhibited which
results in a decrease of the color intensity.

Targeting the same glyphosate analyte, Li Haiyin et al. [50] have reported a peroxidase-mimetic 243 nanozyme to develop a portable Paper-based Analytical Device (PAD) allowing for its 244 detection. In this article, the preparation of 2D nanosheet-like V₂O₅ (2D-VONz) with exclusive 245 peroxidase-mimetic activity under optimal reaction conditions is described. Interestingly, the 246 247 activity of 2D-VONz is inhibited by glyphosate, which was then exploited to develop a PAD, on which, glyphosate reduces the activity of 2D-VONz to prevent the catalytic oxidation of 248 3,3',5,5'-tetramethylbenzidine, thus contributing to rapid, naked-eye and portable analysis of 249 250 glyphosate using a smartphone. Another highly efficient biosensor, based on an enzymemediated dephosphorylation nanozyme, has been developed for the rapid, specific, and 251 sensitive detection of paraoxon [51]. This biosensor utilizes a novel CeO2@N-doped carbon 252 (CeO2@NC) nanozyme [98]. The mechanism underlying the degradation of phosphotriesters, 253 catalyzed by CeO2@NC, involves Ce(IV)/Ce(III) species acting as active sites for the 254 255 polarization and hydrolysis of phosphoester bonds. Additionally, the N-doped carbon (NC) material acts as synergistic sites, facilitating the adsorption of the paraoxon substrate and 256 promoting the hydrolysis process. The characteristic properties of this biosensor are presented 257 258 in Table 2 that shows different types of enzyme-based biosensors developed against pesticides Different types of enzyme-based biosensors developed against pesticides are described in Table 259 2. 260

261 I.2.3. Chemilumiscence Assay
262 Chang et al. [52] introduced a highly sensitive nanozyme chemiluminescence-based (CL)
263 biosensor for the specific detection of glyphosate. The biosensor utilizes a porous hydroxy

264	zirconium oxide nanozyme (ZrOX-OH) obtained through a straightforward alkali solution
265	treatment of UIO66.ZrOX-OH which demonstrates remarkable phosphatase-like activity,
266	enabling the dephosphorylation of 3-(2'-spiroadamantyl)-4 methoxy-4-(3'-
267	phosphoryloxyphenyl)-1,2-dioxetane (AMPPD), which results in the generation of an intense
268	CL signals. Notably, the phosphatase-like ZrO _X -OH exhibited a distinctive response to
269	glyphosate due to the unique interaction between the carboxyl group of glyphosates and the
270	surface hydroxyl group, leading to the development of a CL biosensor for the direct and
271	selective detection of glyphosate without the need for bio-enzymes.

Pesticide	Detection technique	Enzyme used	LOD	Validation with real samples	Reference
Malathion, chlorpyrifos, monocrotophos and endosulfan	DPV	AChE	33 ng L ⁻¹ , 35 ng L ⁻¹ 223.2 ng L ⁻¹ ¹ , and 4 μg L ⁻¹	Milk	Chauhan and Pundir (2011) [37]
Glyphosate	Potentiometric assay	Urease	0.5 mg L ⁻¹	Tap water	Vaghela et al (2018) [35]
Carbendazim, chlorpyrifos, dichlorodiphenyltri chloroethane (DDT), dinocap, and ethion	CA and CV	Glutathione-S- transferase	2 μg L ⁻¹ , 60 μg L ⁻¹ , 40 μg L ⁻¹ , 50 μg L ⁻¹ , and 100 μg L ⁻¹	Potatos	Borah et al (2018) [46]
Chlorpyrifos	High open circuit voltage (photoelectroch emical enzymatic fuel cell)	AChE	0.012 µg L ⁻	Not mentioned (PB)	Gai et al (2018) [39]
Omethoate	DPV	AChE	0.36 ng L ⁻¹	Cabbage and cucumber	Ma et al (2018) [38]
Paraoxon	DPV	2D MnO ₂ sheets/AChE nanozyme	25 ng L ⁻¹	Pakchoi	Wu et al. [45]
Paraoxon	СА	AChE	13 ng L ⁻¹	Apple and pakchoi	Zhao et al (2017) [47]

Table 2 Enzyme-based biosensors developed for pesticides detection.

Paraoxon, 2.4- dichlorophenoxyac etic acid, and atrazine	СА	BChE, alkaline phosphatase, and tyrosinase	2 μg L ⁻¹ , 50 μg L ⁻¹ , and 10 μg L ⁻¹	River water	Arduini et al (2019) [28]
Malathion	CV and EIS	AChE	0.39 μg L ⁻¹	Cabbage and carrots	Li et al (2020) [53]
Nitrofen	DPV	Candida Rugosa Lipase	7.38 μg L ⁻¹	Apricot	Cheng et al (2021) [54]
Ethyl paraoxon	CV	AChE-ChOx	0.46 ng L ⁻¹	Pakchoi, cabbage and lettuce	Yang et al (2021) [55]
Paraoxon	DPV	AChE	1.7 μg L ⁻¹	Apple and Eggplant	Li et al (2020) [56]
Chlorpyrifos	DPV	AChE	20 ng L ⁻¹	Cabbage and spinach	Wang et al (2016)[57]
Chlorpyrifos	CV	AChE	50 ng L ⁻¹	Cabbage, rape and lettuce	Chen et al (2017) [58]
Malathion and methyl parathion	CV	AChE	0.31 pg L ⁻¹ and 0.188 pg L ⁻¹	Tap water and Chinese cabbage	Jian et al (2019) [59]
Fenthion	CV and EIS	AChE	100 μg L ⁻¹	Cabbage juice	Cui et al (2019) [60]
Phosmet	CV and EIS	Tribolium castaneum acetylcholinester ase	1.14 μg L ⁻¹	Wheat flour	Bial et al (2021) [61]
11 organophosphorus and methomyl	DPV and EIS	AChE	Organopho sphorus: 19 – 77 ng L ⁻¹ Methomyl: 81 ng L ⁻¹	Trichlorfon and dichlorvos: Apple and cabbage	Zhao et al (2021) [62]
Paraxon	CV	AChE	1.4 μg L ⁻¹	Vegetable leaves	Jia et al (2020) [63]
Paraxon	CV	AChE	4 µg L ⁻¹	Chinese chives and cabbage	Chen et al (2020) [64]
Paraoxon, dichlorvos, malathion and triazophos	Fluorescence	AChE and ChOx	0.44 pg L ⁻¹ , 16.6 pg L ⁻¹ , 76 μg L ⁻¹ and 3 ng L ⁻¹	ND	Korram et al (2020) [65]
Pirimicarb, dichlorvos and carbaryl	Fluorescence	AChE	50 μg L ⁻¹ , 10 μg L ⁻¹ and 10 μg L ⁻¹	Pirimicarb: Lettuce, choy and rice	Apilux et al (2017)[66]

	Ethylparathion	thion Fluorescence AChE 0.		0.7 ng L ⁻¹	Tap water, soil water, apple and orange juice	Sharma et al (2021) [67]
	Parathionmethyl, monocrotophos and dimethoate	Fluorescence	cence AChE $\begin{array}{c} 0.67 \text{ ng } \text{L}^{-1}, \\ 23 \text{ ng } \text{L}^{-1} \\ \text{and } 67 \text{ ng } \\ \text{L}^{-1} \end{array}$		Apple, cucumber and capsicum	Long et al (2015) [41]
	Carbaryl, ethodan, endosulfan and acephate	Colorimetry and fluorescence	AChE	0.4 to 3 µg L^{-1}	River water	Luo et al (2018) [68]
	Glyphosate	Fluorescence Exonuclease I		/	Drinking water	Berkal et al (2023) [36]
	Glyphosate	Colorimetry	peroxidase- mimetic nanozyme/ 2D-VONz	/	/	Li Haiyin et al. (2023) [50]
	Paraxon	Colorimetry	Phosphatases- mimetic nanozyme/CeO 2@NC	0.82 mg L ⁻¹	Garlic chives	Gai et al. (2022) [51]
Glyphosate		Chemiluminisc ence	Phosphatases- mimetic nanozyme /ZrO _X -OH	55.8 mg L ⁻¹	Cabbage juice	Chang et al. [52] (2023)

Upon reviewing **Table 2**, notable disparities in the Limit of Detection (LOD) become evident among biosensors utilizing the same bioreceptor and targeting the identical pesticide. This variability predominantly stems from the choice of the detection technique employed. Certain detection methods prove more accurate to detect pesticides than others. Furthermore, additional factors, such as the composition of the analyzed matrix, whether it is a simple medium like water or a more complex one like food or vegetables, also contribute significantly to the observed variation in LOD values.

280 I. Immunosensors

Immunosensors are characterized by the directed and highly selective interaction between a ligand, or antigen (Ag), and its antibody (Ab), immobilized on the transducer surface [69–72]. Once the equilibrium is reached, the ratio of bound-to-free antigen is quantitatively related to the global amount of ligand. Unlike AChE inhibition based biosensors, immunosensors have the advantage of being specific [73]. Several immunosensors have been developed forpesticides detection, based essentially on optical and electrochemical transduction methods.

Given the relatively small size of pesticides, competitive assay schemes are frequently utilized for analyzing pesticides with immunosensors [74]. Based on the detection principle, these immunosensors can be classified into two main categories, namely optical and electrochemical immunosensors. In contrast, electrochemical immunosensors utilize changes in current or potential to detect pesticides, offering several advantages, such as rapid detection, portability, and low cost.

293

I.1. Optical immunosensors

Optical immunosensors rely on the specific binding interactions between an antibody or antigen and its corresponding analyte, resulting in a detectable signal change that can be measured using optical techniques such as absorbance, fluorescence, or surface plasmon resonance. These immunosensors offer several benefits, including high sensitivity, selectivity, real-time detection, and the ability to be easily miniaturized for use in portable devices.

299

I.1.1. Colorimetric immunosensors

Colorimetric immunoassays have several advantages over conventional methods for pesticides 300 analysis, such as rapid detection, ease of use, suitability for high-throughput analysis with 301 limited technical requirements [75]. Liu and coworkers [76] have developed a colorimetric 302 immunochromatographic strip (ICA) using a monoclonal antibody (mAb) for the detection of 303 304 dicofol (Figure 4(A)), an organo-insecticide widely spread on vegetables, fruits, teas, ornamental plants, and field crops [77]. The ICA displayed high sensitivity with a limit of 305 detection of 50 μ g L⁻¹ with both apple and cucumber, and the cross-reactivity test revealed a 306 good specificity for dicofol. These results were consistent with LC-MS and immunocapture 307 ELISA, suggesting that the ICA method is reliable and practical for detecting dicofol in fruits 308 and vegetables. 309

Fig. 4



310

(A) Schematic of the ICA strip (a) and interpretation of the test results (b). Reprinted from [76], 311 with permission from Royal Society of Chemistry. (B) Novel immunoassay containing 312 glyphosate-double DNA-gold nanoparticles based on competitive inhibition reaction. 313 Reprinted (adapted) from [78], with permission from American Chemical Society. (C) 314 Preparation of mAb-immobilized sensor chip for direct detection of triazophos and real-time 315 SPR sensorgram for association and dissociation of the immunocomplex. Reprinted (adapted) 316 from [79], with permission from Elsevier. (D) Schematic of the graphene-based screen-printed 317 immunosensor for parathion. Reprinted (adapted) from [80], with permission from Elsevier. 318

319

I.1.2. Fluorescence-based immunosensors

Gonzalez-Martinez and coworkers [81] have developed a fluorescent immunosensor based on 320 an immunocomplex capture assay protocol to detect glyphosate. The described immunosensor 321 is fully automated and performs on-line analyte derivatization prior to the assay. It utilizes a 322 highly selective anti-glyphosate serum, a glyphosate peroxidase enzyme tracer, and fluorescent 323 detection system for high sensitivity and accuracy. Its specific and sensitive detection reaches 324 a limit of detection of 0.021 µg L⁻¹, which is lower than the concentration limit set by the 325 European council directive [82]. Lee et al. [78] have also described a new method for detecting 326 327 glyphosate using Co–B/SiO₂/dye nanoparticles in a water-in-oil microemulsion (Figure 4(B)). The nanoparticles have been surface-modified to enhance their detection limit. The approach 328 utilized fluorescence magnetic nanoparticles (FMP) and a glyphosate antibody. With this 329

method, a limit of detection of 45.6 ng L⁻¹ was reached, with a linear correlation in the range of $0.169 \ \mu g \ L^{-1} - 1.69 \ m g \ L^{-1}$.

332 I.1.3. Reflectometric interference spectroscopy immunosensors

Reflectometric interference spectroscopy presents several advantages over classical optical 333 techniques such as simple instrumentation, no-need for optical alignment and low cost of the 334 sensing element. Koukouvinos et al. [83] have developed a white light reflectance spectroscopy 335 336 (WLRS) based immunosensor for the fast, real time and label-free simultaneous detection of chlorpyrifos, imazalil and thiabendazole pesticides in drinking water and wine samples. The 337 biosensor allowed the entire detection of those pesticides within 10 min. The accuracy of the 338 measurements was evaluated through recovery experiments and comparison of the results with 339 validated liquid chromatography coupled with tandem mass spectrometry (LC-MS/MS). 340 Recovery values ranged from 86 to 116%, and the results were in good agreement with LC-341 MS/MS. 342

343 I.1.4. Surface Plasmon Resonance-(SPR) based immunosensors

SPR is an optical detection platform widely used for the detection of large molecules, with 344 which the high mass of the analyte and the use of a sandwich immunoassay format provide a 345 346 high signal and thus the desired sensitivity [84, 85]. In contrast and due to the low change of refractive index induced by the binding of these analytes to the sensor surface, small molecular 347 compounds such as pesticides present at low concentrations are very difficult to be directly 348 349 detected by traditional SPR immunosensors [86]. Recent improvements in SPR devices, including lower-noise valves and improved microfluidics with more efficient vacuum pumps, 350 have reduced the overall noise of these systems and thus improved sensitivity and reliability 351 [87, 88]. Owing to these improvements, SPR immunosensors could be employed for pesticide 352 monitoring. Guo et al. [79] have described an SPR immunosensor for the detection of an 353 organophosphate pesticide, triazophos (Figure 4(C)). They immobilized two anti-triazophos 354 monoclonal antibodies (mAbs) on the sensor chip, and characterized them by SPR-based kinetic 355

analysis. The mAb, characterized by a relatively slow dissociation rate, could be used for the
development of a non-competitive SPR-immunosensor for direct monitoring of triazophos
residue in environmental and agricultural samples.

359

I.2. Electrochemical immunosensors:

Electrochemical immunosensors are interesting alternatives to classical chromatographic methods for pesticide analysis. Owing to their ability to detect specific targets with high sensitivity, they are promising tools for rapid and on-site analysis of pesticides [89–91].

- I.2.1. Cyclic voltammetry
 Talan et al. [92] have described a highly sensitive fluorine doped tin-oxide (FTO) based
 electrochemical immunosensor to detect chlorpyriphos using gold nanoparticles (AuNPs) and
 anti-chlorpyrifos antibodies (Chl-Ab). The developed biosensor allowed, via cyclic
 voltammetry measurements, the successful detection of chlorpyrifos with very -high sensitivity
 ranging from 1 fM to 1 μM with a LOD of 10 fM.
- 369 I.2.2. Chronoamperometric assay

Bettazzi et al. [93] developed an electrochemical competitive immunoassay that utilizes 370 371 antibody-modified magnetic particles. The assay works by employing a Horseradish Peroxidase (HRP)-conjugated glyphosate tracer and anti-glyphosate IgG-modified magnetic beads (MBs) 372 for detection. A screen-printed electrochemical cell has been used to detect glyphosate. The 373 calibration curve demonstrated a linear concentration range of 0 - 10000 ng L⁻¹ with a LOD of 374 5 ng L⁻¹ and LOQ of 30 ng L⁻¹ which is well below the EU legislative recommendations. 375 Furthermore, the developed biosensor showed a good applicability for the analysis of real 376 samples, namely spiked beer samples. 377

378

I.2.3. Impedance spectroscopy

A graphene-based immunosensor was developed and used to detect parathion [80] (**Figure** 4(**D**)). The process involved modifying screen-printed carbon electrodes with graphene sheets and their functionalization with 2-aminobenzyl amine before bio-interfacing with antiparathion antibodies. The biosensor demonstrated a broad linear range of detection (0.1–1000 ng L⁻¹) and a very low limit of detection (52 pg L⁻¹) with high selectivity towards parathion.
The biosensor was also successfully used to detect parathion in real samples such as tomato and
carrot and was cross-calibrated against high performance liquid chromatography (HPLC) to
confirm its viability. Table 3 provides a summary of the various types of immunosensors
developed for the detection of pesticides.

388 Table 3 Immunosensors developed against pesticides

Pesticide	Detection technique	LOD	Validity for real samples	Reference
Parathion, methyl- parathion, and fenitrothion	ELISA	2.5 μg L ⁻¹ for parathion and methyl-parathion 5 μg L ⁻¹ for fenitrothion	Tap water	Jiao et al. (2018) [94]
Glyphosate	Fluorometric assay	0.021 μg L ⁻¹	Water and soil	Gonzalez- Martinez et al. (2005) [81]
Glyphosate	Fluorometric assay	10 µg L ⁻¹	Not mentioned (PBS buffer)	Lee et al. (2010) [78]
Acetochlor and fenpropathrin	Colorimetry	0.6 μg L ⁻¹ and 0.24 μg L ⁻¹	Corn, apple and cabbage	Cheng et al (2019) [95]
Glyphosate	Chronoamperometric assay	5 ng L ⁻¹	Commercial beer	Betazzi et al. (2018) [93]
Chlorpyrifos	DPV and CV	3.5 pg L ⁻¹	Apple, cabbage and pomegranate	Talan et al. (2018) [92]
Thiazophos	SPR	96 ng L ⁻¹	Environmental water	Guo et al. (2018) [79]
Chlorpyrifos, amazalil, and thiabendazol	WLRS	0.6 μg L ⁻¹ for chlorpyrifos and imazalil 0.8 μg L ⁻¹ for thiabendazol	Drinking water and wine	Koukouvinos et al. (2017) [83]
Chlorpyrifos	EIS	70 pg L ⁻¹	Chinese cabbage and lettuce	Hou et al (2020) [96]
Malathion	DPV	0.33 pg L ⁻¹	Lettuce	Kaur et al (2021) [97]
Endosulfan and paraoxon	CV and SWV	$\begin{array}{c} 0.05 \ \mu g \ L^{-1} \ and \ 2 \\ \mu g \ L^{-1} \end{array}$	Environmental water	Liu et al (2014) [98]
Paraquat	Amperometry	1.4 µg L ⁻¹	Potatos	Valera et al (2014) [99]
Parathion	EIS	52 pg L ⁻¹	Tomato and carrot	Mehta et al (2016) [80]

389 II. Aptasensors

The development of the Systematic Evolution of Ligands by Exponential enrichment (SELEX) technique has led to the discovery of aptamers, oligonucleotide sequences composed of nucleotide bases (Adenine, Thymine, Cytosine, Guanine), capable of complexing a target with very high affinity and specificity. Aptamers offer many advantages over antibodies (**Table 4**) and are specific to a wide variety of targets: ions [100–102], proteins [103], small molecules [104–106], and cells [107, 108]. These characteristics make aptamers excellent candidates to
replace antibodies in various fields, more particularly environmental analysis that requires
stable and robust sensors.

398	Table 4 Elevating the Distinctive Attributes of Aptamers over Antibodies: A Comparative
399	Analysis [109].

Properties	Aptamers	Antibodies
Stability	Withstand repeated rounds of	Easily denatured.
	denaturation/renaturation. Temperature	Temperature sensitive and require
	resistant: stable at room temperature.	refrigeration to avoid denaturation.
	Long shelf life (several years). Can be	Limited shelf life.
	lyophilized.	Must be refrigerated for storage and transport.
	Degradable by nucleases.	Degradable by proteases.
	Resistant to proteases.	Resistant to nucleases.
Synthesis	In vitro SELEX takes only 2–8 weeks	Produced in vivo
	No batch-to-batch variation Cheap to	More than 6 months
	synthesize	Batch-to-batch variations
		Laborious and expensive
Target potential	From ions and small molecules to whole cells	Targets must cause a strong immune response
	and live animals	for antibodies to be produced
Size	Small molecules	Relatively large by comparison
Modifiability	Aptamers can readily and easily be modified without affinity loss	Modifications often lead to reduced activity
Affinity	High and increased in multivalent aptamers	Dependent on the number of epitopes on the antigen.
Specificity	Single point mutations identifiable	Different antibodies might bind the same antigen
Tissue uptake/kidney	Fast	Slow
filtration		

Conventional SELEX methods are however not compatible with small molecules because of 400 401 the immobilization step that is not only complicated [110], but leaves weak functional groups exposed to the oligonucleotides thus reducing the chance of selecting aptamers. In 2005, 402 Stoltenburg et al. reported a method for selecting fluorophore-labeled aptamers that involves 403 404 the grafting of oligonucleotides onto magnetic beads (FluMag-SELEX) [111], which subsequently furthered Capture-SELEX, a method suitable for small molecule aptamers 405 406 selection. Unlike conventional SELEX, the immobilization in Capture-SELEX is done with the oligonucleotides and not with the target molecules, which allows the preservation of all 407 408 functional groups of these molecules thus increasing the chance of selecting aptamers. The operating principle of the Capture-SELEX can be summarized in the following steps (Figure 409 5): 410

- 411 1. Immobilization: the capture oligonucleotide (DNA complementary to the library) is
 412 immobilized on the solid matrix (magnetic beads for example).
- 413 2. Hybridization: the oligonucleotide library is hybridized with the immobilized capture414 oligonucleotides.
- 415 3. Incubation: the target is incubated with the oligonucleotide library.

- 416 4. Elution: the sequences with an affinity towards the target will detach from the capture417 oligonucleotide and go into solution.
- 418 5. Amplification: the recovered sequences are enriched by Polymerase Chain Reaction419 (PCR) for the next round.
- 420 Similar to conventional SELEX, candidate aptamers are sequenced and cloned at the end of
- 421 the Capture-SELEX (6 to 20 rounds) to identify the aptamers that have the desired

422 properties.



423

424 Schematic illustration of Capture-SELEX using magnetic beads. Reprinted from [112], with425 permission from American Chemical Society.

426

II.1. Optical aptasensors

Optical aptasensors have become a well-established technology in the field of bioanalysis, utilizing aptamers as recognition elements to directly target and identify substances. These sensors utilize various light sources, including fluorescence, colorimetry, SERS, etc., and convert signals in the form of UV, visible, and IR radiation into different data formats. Compared to other detection methods, optical aptasensors offer several advantages, such as low cost, repeatable use, high specificity, simple sample preparation, minimal interference, and high 433 accuracy. These characteristics make optical aptasensors particularly suitable for on-site434 detection applications [113].

II.1.1. **Colorimetric aptasensors** 435 436 The primary benefit of colorimetric sensing is its straightforwardness and ease of implementation. Colorimetric aptasensors have been extensively employed for detecting 437 pesticides in real samples due to the convenience of visual observation. However, a significant 438 439 challenge in designing colorimetric aptasensors relies on converting the response change into a 440 noticeable color change. Gold (Au) and silver (Ag) nanoparticles are the most commonly used probes in colorimetric assays, due to their ability to enhance surface plasmon resonance and 441 442 thus produce robust signals as monitored by colorimetry (Figure 6(A)) [114, 115].



443

444 Comparative schematic diagram of (A) colorimetric aptasensor, (B) FRET aptasensor and (C)

electrochemical aptasensor. (B) Reprinted from [116], with permission from ScienceDirect.

446 II.1.2. Fluorescence-based aptasensors

Fluorescence-based aptasensors rely on the correlation between changes in fluorescence 447 intensity and the concentration of the target upon aptamer binding. Fluorescence changes are 448 typically achieved through the introduction of a fluorescence quenching or fluorescence 449 resonance energy transfer agent (FRET, Figure 6(B)). FRET occurs between two fluorophores 450 at an appropriate distance and can utilize various materials as fluorescent donors, including 451 dyes, quantum dots, metal nanoparticles, carbon dots, and carbon nanomaterials. 452 Nanomaterials, such as graphene oxide, gold nanoparticles, carbon nanotubes, and 453 nanocomposites, are also used as well-known fluorescence quenchers [117]. 454

To create efficient aptasensors with high sensitivity, high throughput, and multiplexing 455 capabilities, various combinations of nanomaterials are employed as fluorescent donors and 456 receptors in sensing assays. Fluorescence detection relies on switching the fluorescence signals 457 on and off to indicate the presence of analytes [118, 119]. In most cases, fluorescence signals 458 are in the "signal-off" state until the target is present and the blocking of signals is lifted, leading 459 to a "signal-on" response. Fluorescence signals can be amplified using a variety of 460 nanomaterials such as gold and silver nanoparticles, known for their superior performance [120, 461 462 121]. Su et al. proposed a method for detecting carbendazim (CBZ) in water using a fluorescent aptasensor [122]. The aptasensor used a CBZ-specific aptamer as a sensing probe, gold 463 nanoparticles (AuNPs) and Rhodamine B (RhoB) as an indicator. In the absence of CBZ, the 464 aptamer wrapped around the AuNPs and kept them dispersed in a solution. However, in the 465 presence of CBZ, the aptamer forms a complex with CBZ, leaving the AuNPs which aggregate 466 in a NaCl solution. The concentration of CBZ was determined by measuring the fluorescence 467 intensity. The method had a wide linear range from 2.33 to 800 nM and a LOD of 2.33 nM. The 468 fluorescent aptasensor showed potential for use in detecting CBZ in aquatic environments. 469

470

II.1.3. Luminescence-based aptasensors

Common fluorescence techniques that rely on organic dyes are vulnerable to interference from 471 naturally occurring fluorescent substances in the environment [123]. Time-resolved emission 472 spectroscopy (TRES) is a useful method for discriminating between fluorophores with similar 473 emission spectra but varying decay times. TRES records long-lived phosphorescence once the 474 initial short-lived fluorescence background has faded [124-127]. Therefore, luminescent 475 transition metal complexes, such as iridium(III) complexes, have become preferred for TRES 476 due to their excellent optical characteristics and long-lived phosphorescence [128]. Chen et al. 477 [129] have reported on a luminescent G-quadruplex-derived aptasensor for the monitoring of 478 479 glyphosate using a phosphorescent iridium(III)-based probe. The developed platform displayed excellent sensitivity with good selectivity for glyphosate with a limit of detection of 4.46 μ g L⁻ 480 ¹ in Tris-buffer, and 13.4 µg L⁻¹ in Tris-buffer containing soybean extract to simulate a complex 481 482 environment.

483

II.1.4. Phosphorescence aptasensors

Persistent luminescence nanorods (PLNRs) have become a subject of extensive research owing 484 to their unique afterglow or phosphorescence characteristics, which unlike fluorophores allow 485 for reduced or even eliminated background luminescence. These nanorods exhibit persistent 486 luminescence, which refers to the emission of light after the cessation of the excitation source. 487 Wang et al. have developed an aptamer-based colorimetric-phosphorescence assay for the 488 detection of isocarbophos [130]. The colorimetric assay usedAuNPs that were aggregated by 489 490 competitive binding of the aptamer between isocarbophos and AuNPs at high salt concentration. The addition of persistent luminescence nanorods (PLNRs) to the system 491 resulted in phosphorescence that was sensitive to the concentration of isocarbophos due to the 492 inner filter effect between PLNRs and AuNPs. The assay showed good linearity within the 493 range of 50-500 μ g L⁻¹ and 5-160 μ g L⁻¹, with a limit of detection of 7.1 μ g L⁻¹ and 0.54 μ g L⁻¹ 494 in colorimetry and phosphorescence mode, respectively. The method was successfully 495

demonstrated for food analysis with the detection of isocarbophos residues in vegetables,showing both sensitivity and selectivity.

498

II.1.5. SERS-based aptasensors

Surface-Enhanced Raman spectroscopy (SERS) is a powerful spectroscopic technique that 499 500 leverages plasmonic metal nanostructures for the detection of trace-level chemicals with high sensitivity and selectivity [131]. SERS has been widely adopted in analytical chemistry [132], 501 biochemical analysis [133], and environmental sensing [134, 135], including the detection of 502 503 harmful agrochemicals, such as pesticides [136–138]. However, the lack of selectivity when the target molecule is mixed with other molecules in a complex matrix and susceptibility to 504 interference from various ambient influences can limit SERS detection techniques. Therefore, 505 sample pretreatment methods, such as purification and extraction, are critical for examining 506 environmental samples. The use of aptamers, specific biomolecules that can enhance the 507 508 selectivity of target analysis when combined with SERS, can address these limitations. Recently, Kamkrua et al. [139] developed a SERS aptasensor using gold nanoparticles to detect 509 parquat, a commonly used herbicide. The platform used AuNPs-SERS substrates modified with 510 511 a thiol-modified aptamer as bioreceptor to selectively bind to parquat molecules. The aptamermodified SERS substrate showed improved sensitivity and selectivity compared to SERS 512 substrates without aptamer modifications, with a limit of detection of $0.10 \pm 0.03 \mu$ M. The 513 platform was evaluated using natural water samples and demonstrated good stability against 514 interferences from other similar herbicides and insecticides present along with parquat in water. 515

516

II.2. Electrochemical aptasensors

Electrochemical aptasensors produce an electrochemical signal upon the capture of a particular target substance, achieved through the use of aptamers. The design of electrochemical aptasensors involves three crucial steps, including the utilization of nanomaterials to amplify the signal, binding of the aptamer to the sensor surface, and the detection process itself (**Figure**

6(C)). Covalent bond formation is the most commonly employed method for aptamer 521 522 immobilization onto the surface in these types of sensors [140]. Numerous electrochemical aptasensors have been reported for the identification of proteins, which are large molecules 523 producing significant variation in electrochemical signal upon binding to the aptamer, thereby 524 facilitating their detection. However, detecting pesticides which are small molecules requires a 525 notable change in aptamer conformation to enable their detection by electrochemical platforms. 526 527 To overcome this challenge, multiple approaches have been adopted, one of which is the use of oligonucleotide switching structures generated through the Capture-SELEX method that has 528 proven to be more suitable for detecting pesticides due to the aptamers selection process: a 529 530 significant conformational change occurs upon binding to the target [141]. This conformational change leads to an improved ability to detect and quantify small molecule pesticides using 531 electrochemical aptasensors. In Table 5, various aptasensors that were developed for the 532 533 detection of pesticides have been listed.

 Table 5 Recent advances in aptasensors developed against pesticides.

Target	Detection technique	Aptamer sequence	LOD	Dynamic range	Validity for real samples	Reference
Acetamiprid	EIS	Acetamiprid: 5'-(SH)-(CH2)6- TGTAATTTGTCTGCAGCGGTTCTTGATCGCTGA CACCATATTATGAAGA-[FITC]-3'	Acetamiprid: 8.9 ng L ⁻¹	ND	Buffer	Madianos et al (2018)
Atrazine		Atrazine: 5'-(SH)-(CH2)6- TACTGTTTGCACTGGCGGATTTAGCCAGTCAGT G-[FITC]-3'	Atrazine: 1.3 ng L ⁻¹			[142]
Acetamiprid	EIC	Acetamiprid: 5'-(SH)-(CH2)6- TGTAATTTGTCTGCAGCGGTTCTTGATCGCTGA CACCATATTATGAAGA-[Flc]-3'	Acetamiprid: 0.2 ng L ⁻¹	Acetamiprid: 2.2 ng L ⁻¹ - 22.2 μg L ⁻¹	Toursuster	Madianos et
Atrazine	EIS	Atrazine: 5'-(SH)-(CH2)6- TACTGTTTGCACTGGCGGATTTAGCCAGTCAGT G-[Flc]-3'	Atrazine: 2.1 ng L ⁻¹	Atrazine: 21.5 ng L ⁻¹ – 215.7 μg L ⁻¹	Tap water	[143]
Atrazine	Photoelectroch emical assay	5'-SH-(CH2)6-TGT-ACC-GTC-TGA-GCG-ATT- CGT-ACG-AAC-GGCTTT-GTA-CTG-TTT-GCA- CTG-GCG-GAT-TTA-GCC-AGT-CAG-TGT-TAA- GGA-GTG-C-3'	0.21 ng L ⁻¹	0.5 ng L ⁻¹ – 0.11 μg L ⁻¹	Lake water, agricultural wastewater and sewage water	Fan et al (2021) [144]
Carbofuran	DPV	5'- CACCTGGGGGGAGTATTGCGGAGGAAAGAGAAC ACTGGGGCAGATATGGGCCAGCAGGTC- (CH2)3-SH-3'	14.8 μg L ⁻¹	44 ng L ⁻¹ -11 μg L ⁻¹	Vegetables and fruits	Li et al (2018) [145]
Chlorpyrifos	DPV	5'- CCTGCCACGCTCCGCAAGCTTAGGGTTACGCCT GCAGCGATTCTTGATCGCGCTGCTGGTAATCCT TCTTTAAGCTTGGCACCCGCATCGT-3'	70 ng L ⁻¹	0.1 μg L ⁻¹ - 150 μg L ⁻¹	Apple and celery cabbage	Xu et al (2018) [146]

Atrazine	EIS	5' -TGT ACC GTC TGA GCG ATT CGT ACG AAC GGC TTT GTA CTG TTT GCA CTG GCG GAT TTA GCC AGT CAG TGT TAA GGA GTG C-3'	0.67 ng L^{-1}	1 ng L ⁻¹ -50 μg L ⁻¹	River and tap water	Zhu et al (2021) [147]
Atrazine	Photoelectroch emical assay	5'-TGT-ACC-GTC-TGA-GCG-ATT-CGT-ACG-AAC- GGC-TTT-GTA-CTG-TTT-GCA-CTG-GCG-GAT- TTA-GCC-AGT-CAG-TGT-TAA-GGA-GTG-C-3'	2.6 pg L ⁻¹	10.7 pg L ⁻¹ – 64.7 ng L ⁻¹	Water	Sun et al (2019) [148]
Atrazine	DPV	5'-SH-(CH2)6-TGT-ACC-GTC-TGA-GCG-ATT- CGT-ACG-AAC-GGCTTT-GTA-CTG-TTT-GCA- CTG-GCG-GAT-TTA-GCC-AGT-CAG-TGT- TAAGGA-GTG-C-3'	21.5 pg L ⁻¹	53.9 pg L ⁻¹ – 53.7 ng L ⁻¹	Lake and river water	Fan et al (2019) [149]
Atrazine	Linear Sweep Voltametry (LSV)	5'-HS-(CH2)6-TGT-ACC-GTC-TGA-GCG-ATT- CGT-ACG-AAC-GGCTTT-GTA-CTG-TTT-GCA- CTG-GCG-GAT-TTA-GCC-AGT-CAG-TGT- TAAGGA-GTG-C-3'	1.6 μg L ⁻¹	ND	Environmental water	Wang et al (2020) [150]
Atrazine	Adsorption spectroscopy (ATR- SEIRAS)	5'-HS-(CH2)6-TGT-ACC-GTC-TGA-GCG-ATT- CGT-ACG-AAC-GGC-TTT-GTA-CTG-TTT-GCA- CTG-GCG-GAT-TTA-GCC-AGT-CAG-TGT-TAA- GGA-GT7G-C-3'	0.23 μg L ⁻¹	ND	ND	Sun et al (2021) [151]
Atrazine	Ultrafiltration system	5'-/FAM/- TACTGTTTGCACTGGCGGATTTAGCCAGTCAGT G-3'	ND	ND	ND	Romero- Reyes and Heemstra (2021) [152]
Atrazine	Fluorescence	5'-TTT-TTT-TTT-TTA-CTG-TTT-GCA-CTG-GCG- GAT-TTA-GCC-AGT-CAG-TG-3'	2 ng L ⁻¹	ND	River simple	Yao et al (2021) [153]
Atrazine	SERS	5'-(SH)-TACTG TTTGC ACTGG CGGAT TTAGC CAGTC AGTG-3'	$0.14 \ \mu g \ L^{-1}$	0.22 μg L ⁻¹ – 10.7 μg L ⁻¹	Cherry tomato and grape	Wei et al (2020) [154]
Carbendazim	Fluorescence	5'- CGACACAGCGGAGGCCACCCGCCCACCAGCCC CTGCAGCTCCTGTACCTGTGTGTGTG-3'	$0.44 \ \mu g \ L^{-1}$	$0.44 \ \mu g \ L^{-1} - 0.15 \ mg \ L^{-1}$	Water	Su et al (2020) [122]
Chloramphen icol	Colorimetry	5'- ACTTCAGTGAGTTGTCCCACGGTCGGCGAGTC GGTGGTAG-Biotin-3' and atibody	0.19 μg L ⁻¹ in milk 0.23 μg L ⁻¹ in serum	ND	Milk and mouse serum	Abnous et al (2016) [155]

Chlorpyrifos	Colorimetry	5'- CCTGCCACGCTCCGCAAGCTTAGGGTTACGCCT GCAGCGATTCTGATCGCGCTGCTGGTAATCCTT CTTTAAGCTTGGCACCCGCATCGT-3'	11.3 mg L ⁻¹	10 mg L ⁻¹ - 200 mg L ⁻¹	River water	Weerathunge et al (2019) [156]
Chlorpyrifos and diazinon and	Fluorescence	Chlorpyrifos aptamer 5'- CCTGCCACGCTCCGCAAGCTTAGGGTTACGCCT GCAGCGATTCTTGATCGCGCTGCTGGTAATCCT TCTTTAAGCTTGGCACCCGCATCGT-3'	0.73 ng L ⁻¹ and 6.7 ng L ⁻¹	ND	12 vegetables and fruits	Cheng et al (2018) [157]
Diazinon	Fluorescence	5'-NH2-C6- ATCCGTCACACCTGCTCTAATATAGAGGTATTG CTCTTGGACAAGGTACAGGGATGGTGTTGGCT CCCGTAT-3'	23 ng L ⁻¹	0.05 μg L ⁻¹ - 500 μg L ⁻¹	Tea, apple and tap water	Ron et al (2020) [158]
Glyphosate	Fluorescence	5'-CGC ATT CAG GAT TGC ATG ATT GCC AAA AAA AAA A-NH2-3'	10 μg L ⁻¹	10 μg L ⁻¹ - 100 mg L ⁻¹	PBS buffer	Lee et al (2010) [78]
Glyphosate	SERS	5'-TGC TAG ACG ATA TTC GTC CAT CCG AGC CCG TGG CGG GTC TTA GGA CTC TGC GGG CTT CGC GGC GCT GTC AGA CTG AAT ATG TCA-3'	0.34 ng L ⁻¹	0.51 ng L ⁻¹ – 11.8 ng L ⁻¹	Soil	Liu et al (2021) [159]
Glyphosate	Luminiscence	5'- TGCTAGACGATATTCGTCCATCCGAGCCCGTGG CGGGCTTTAGGACTCTGCGGGGCTTCGCGGCGCT GTCAGACTGAATATGTCA-3'	 4.5 μg L⁻¹ in buffer 13.4 μg L⁻¹ in buffer containing soybean extract 	8.5 μg L ⁻¹ – 50.7 μg L ⁻¹	Buffer containing soybean extract	Chen et al (2020) [129]
Glyphosate, malathion and trichlorfon	Fluorescence	5'6-FAM- AGCTTGCTGCAGCGATTCTTGATCGCCACAGA GCT-3'	88.8 ng L ⁻¹ , 195.37 ng L ⁻¹ and 72.2 ng L ⁻¹ ,	0.1 μg L ⁻¹ -10 mg L ⁻¹	Lettuce and carrot	Jiang et al (2020) [160]
Isocarbophos	Fluorescence	5'- ATTCTTGATCGCCACGGTCTGGAAAAAGAGAA GTGGGTAGGGCGGGTTGG–3'	2.9 μg L ⁻¹	2.9 μg L ⁻¹ – 0.14 mg L ⁻¹	Chinese cabbage and apple	Li et al (2018) [161]

Isocarbophos	Phosphorescen ce and colorimetric assays	5'- AGCTTGCTGCAGCGATTCTTGATCGCCACAGA GCT-3'	Phosphoresce nce: 0.54 μg L ⁻¹ Colorimetry: 7.1 μg L ⁻¹	Phosphoresce nce: $5-160$ $\mu g L^{-1}$ Colorimetry: $50-500 \mu g L^{-1}$	Chinese cabbage, brassica rape and lettuce	Wang et al (2019) [130]
Isocarbophos and omethoate	Colorimetry	5'-AAG CTT TTT TGA CTG ACT GCA GCG ATT CTT GAT CGC CAC GGT CTG GAA AAA GAG-3'	0.47 μg L ⁻¹ and 0.35 μg L ⁻¹	50-1000 μg L ⁻¹ and 100-500 μg L ⁻¹	Buffer	Liu et al (2020) [162]
Malathion	Colorimetry	5'- ATCCGTCACACCTGCTCTTATACACAATTGTTT TTCTCTTAACTTCTTGACTGCTGGTGTTGGCTCC CGTAT-3'	0.33 ng L ⁻¹	1.65 ng L ⁻¹ – 3.3 μg L ⁻¹	Human serum	Abnous et al (2018) [163]
Malathion	Fluorescence	5'- ATCCGTCACACCTGCTCTTATACACAATTGTTT TTCTCTTAACTTCTTGACTGCTGGTGTTGGCTCC CGTAT-3'	0.74 ng L ⁻¹	ND	12 vegetables and fruits	Cheng et al (2018) [157]

Conclusion and perspectives

Biosensors have emerged as a promising technology for the detection of pesticides, offering rapid, sensitive, and reliable results. Recent advances in biosensor design, and especially the use of biorecognition elements such as antibodies, enzymes and recently aptamers, have enabled the development of highly sensitive and specific biosensors that can detect a wide range of pesticides with high accuracy.

Despite the numerous advantages of biosensors, several challenges remain to be addressed to unlock their full potential. One key challenge lies in the selectivity of immunosensors and enzyme-based biosensors, which often encounter issues such as cross-interactions among antibodies and a lack of specificity of many enzymes employed in biosensors which are inhibited by organophosphate pesticides, limiting their effectiveness.

Furthermore, biosensors deployed in complex sample matrices face hurdles due to interference from various substances. Mitigating these effects requires the development of effective strategies. Moreover, the limited commercial applications of biosensors call for efforts to scale up production, reduce costs, and comply with regulatory requirements.

To foster wider adoption and successful integration of biosensor technologies, standardized protocols for their development, validation, and manufacturing are essential. Establishing these protocols will ensure consistent performance and facilitate comparability of results across different platforms and laboratories.

Looking to the future, biosensors are expected to play an increasingly important role in the detection and monitoring of pesticides, particularly in the context of environmental and food safety. The integration of biosensors with other technologies, such as smartphone apps and cloud computing, is also expected to further enhance the capabilities of biosensors and expand their applications.

In conclusion, biosensors represent a promising technology for the detection of pesticides, and further research and development in this area is essential to address the challenges and unlock the full potential of this technology for environmental and food safety.

Declarations

No conflicts of interests/Competing interests to declare

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