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FULL PAPER

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Design, synthesis, and antiproliferative effect of 2,9‐bis[4‐ (pyridinylalkylaminomethyl)phenyl]‐1,10‐phenanthroline derivatives on human leukemic cells by targeting G‐quadruplex

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Abstract

Current multiagent chemotherapy regimens have improved the cure rate in acute leukemia patients, but they are highly toxic and poorly efficient in relapsed patients. To improve the treatment approaches, new specific molecules are needed. The G-quadruplexes (G4s), which are noncanonical nucleic acid structures found in specific guanine‐rich DNA or RNA, are involved in many cellular events, including control of gene expression. G4s are considered as targets for the development of anticancer agents. Heterocyclic molecules are well known to target and stabilize G4 structures. Thus, a new series of 2,9‐bis [(substituted‐aminomethyl)phenyl]‐1,10‐phenanthroline derivatives (1a–i) was designed, synthesized, and evaluated against five human myeloid leukemia cell lines (K562, KU812, MV4‐11, HL60, and U937). Their ability to stabilize various oncogene promoter G4 structures (c‐MYC, BCL‐2, and K‐RAS) as well as the telomeric G4 was also determined through the fluorescence resonance energy transfer melting assay and native mass spectrometry. In addition, the more bioactive ligands 1g–i were tested for telomerase activity in HuT78 and MV4‐ 11 protein extracts.

KEYWORDS

1, 10‐phenanthroline, antiproliferative activity, FRET melting, G4 ligands, G‐quadruplex, leukemia

1 | INTRODUCTION

Acute leukemia is a malignant disorder characterized by a blockade of hematopoietic stem cell differentiation. This leads to a high level of immature malignant cells, named blasts. Blast cells accumulate in the bone marrow (BM) and peripheral blood. Acute leukemia prognosis is variable, based on clinical features and leukemia‐specific genetic features including cytogenetic and molecular classifications. The main treatment involves the long‐term use of chemotherapy, but there is a high incidence of relapses in adults, unlike children for whom the prognosis is better. In the last few years, a better understanding of acute leukemia pathophysiology has highlighted several potential therapeutic targets allowing the development of new, more selective therapeutic agents, to improve the overall survival in patients.[1–3]

Telomerase is a ribonucleoprotein reverse transcriptase enzyme that can add telomere repeats to telomere DNA to maintain chromosome stability and integrity. Furthermore, telomerase acts as a key regulator of long‐term proliferation. As 85–90% of cancer cells display telomerase activity, which enables them to proliferate indefinitely and be immortal, telomerase can be considered as an interesting target for cancer chemotherapies. Moreover, most leukemic cells are generally telomerase‐positive and often have shortened telomeres. The rationale for targeting telomeric DNA with drugs is that if the telomere overhang DNA forms higher order structures, such as G‐quadruplexes (G4s), the telomere capping machinery, including the telomerase, is perturbed.[4]

The G4s are single‐stranded guanine‐rich nucleic acid sequences that may fold into noncanonical four‐stranded secondary structures. These structures are formed through π–π stacking of G‐ quartets that involve four guanines organized in a plane via eight Hoogsteen‐type H‐bonding. G4 structures are stabilized in the presence of monovalent cations, such as K^+ or Na^+ . In addition, G4forming motifs have been identified in oncogene promoter regions, bringing to light the therapeutic potential for targeted gene regulation at the transcriptional level. The most studied oncogene promoter G4s include c‐MYC, Bcl‐2, h‐RAS, K‐RAS, and c‐KIT. Thus, both inhibition of telomerase activity through G4 structure stabilization and induction of the DNA damage response through telomere uncapping can prompt a proliferation arrest, resulting in anticancer activity. Hence, small molecules that can target and stabilize G4 structures in the human genome could act as potential and interesting anticancer agents. $[5-7]$

The large planar aromatic surface of a terminal G‐quartet provided a rationale for the design and development of planar G4 ligands such as polyaromatic fused molecules, which include acridines, phenanthroline, quinolone, and quinone.^[8-13] On the basis of these chemical pharmacophores, a number of G4‐binding molecules have been designed and developed as potential anticancer agents during the last two decades, such as SYUIQ‐5, SYUIQ‐FM05, Tz1, quarfloxin, or GTC365 (Figure 1).^[9-14] Many of these heterocyclic ligands are selective for G4 structures over duplex DNA, but the design of a ligand specific for a given G4 structure is still challenging. Among them, APTO‐253, a small‐molecule 2‐indolylimidazole[4,5‐d]phenanthroline derivative (Figure 1), is currently undergoing clinical development for the treatment of acute myeloid leukemia (AML). APTO‐253 was advanced into a phase I clinical trial in patients with relapsed/refractory hematologic malignancies with emphasis on AML. Moreover, APTO‐253 seems to inhibit the expression of the c‐ MYC oncogene, leading to cell cycle arrest and programmed cell death (apoptosis) in human‐derived solid tumor and hematologic cancer cells.^[15-17]

In the course of our work devoted to the discovery of new heterocycles for use in anticancer chemotherapies, [18-23] we previously designed and prepared a series of new substituted 2,9‐bis [(substituted‐aminomethyl)phenyl]‐1,10‐phenanthroline derivatives A (Figure 1), which were designed to bind to DNA G‐ quadruplexes^[24,25] and endowed with interesting activity toward human leukemia cells.^[24] In this context and by considering the biological activities of these previous 2,9‐bis[(substituted‐ aminomethyl)phenyl]-1,10-phenanthrolines against human leukemic cells, we have undertaken the synthesis of a new series of 2,9‐bis[(substituted‐aminomethyl)phenyl]‐1,10‐phenanthroline

derivatives 1a–i. The extended planar aromatic structure of the 1,10‐phenanthroline scaffold provides a basis for efficient π–π stacking interactions with G-quartets.^[26] In this study, we designed and synthesized new phenanthroline derivatives 1a–i substituted with various pyridinylalkylamine chains on the benzyl moieties (Figure 1). We hypothesized that such substitutions at positions 2 and 9 could lead to more potent G4 ligands in terms of selectivity by providing extra aromatic groups. We reasoned that the extended aromatic structures of these pyridinylalkylaminobenzyl groups could provide a basis for further π–π stacking interactions with the G‐quartets and that the length of the amine side chains should influence the stabilizing potential of the ligands. We evaluated ligand‐induced stabilization, specificity, and selectivity of the newly synthesized phenanthroline

FIGURE 1 The structures of SYUIQ‐5, SYUIQ‐FM05, quarfloxin, GTC365, Tz1, APTO‐253, phenanthrolines, A and the newly designed 2,9‐ bis[4‐(pyridinylalkylaminomethyl)phenyl]‐1,10‐phenanthroline derivatives 1a–i

derivatives for various oncogene promoter G4 topologies including c‐MYC, BCL‐2, and K‐RAS through fluorescence resonance energy transfer (FRET) melting experiments. The antiproliferative activity of the obtained derivatives 1a–i was then evaluated in vitro against various hematologic malignant cell lines (K562, KU812, MV4‐11, HL60, and U937). In addition, their cytotoxicity against the HS27a stromal cell line was also evaluated to determine an index of selectivity toward cancer cells. Next, telomerase inhibition ability has been assessed for the more bioactive ligands 1g–i using two cell lines (HuT78 and MV‐4‐11).

2 | RESULTS AND DISCUSSION

2.1 | Chemistry

The reported 2,9-bis[4-(pyridinylalkylaminomethyl)phenyl]-1,10phenanthroline derivatives 1a–i were synthesized starting from the commercially available 2,9‐dichloro‐1,10‐phenanthroline 2 (Scheme 1). The bis‐[2,9‐(formylphenyl)]‐1,10‐phenanthroline 3 was prepared by a direct, double Suzuki–Miyaura cross‐coupling reaction of 2,9‐dichloro‐ 1,10‐phenanthroline 2 with the 4‐formylphenylboronic acid in the

SCHEME 1 General procedure for the preparation of target compounds 1a–i

presence of Pd(PPh₃)₄ as a catalyst and sodium carbonate as the base.[24,25] Reaction of primary substituted pyridinylalkylamines with 3 gave the di‐imines 4a–i, which were reduced into the 2,9‐bis[4‐ (pyridinylalkylaminomethyl)phenyl]‐1,10‐phenanthrolines 1a–i using sodium borohydride in methanol, as previously described.^[24,25] These phenanthroline compounds, 1a–i, were then converted into their ammonium oxalate salts by treatment with oxalic acid in refluxing isopropanol. Table 1 summarizes the physical properties of the 1a–i oxalates.

2.2 | Biological evaluation

The cytotoxic activity of these newly synthesized 2,9‐bis[4‐ (pyridinylalkylaminomethyl)phenyl]‐1,10‐phenanthroline derivatives 1a–i was then evaluated against the K562, KU812, MV4‐11, HL60, and U937 hematologic malignant cell lines with proliferation assays, using derivative APTO‐253 as the reference standard drug.

As noticed in Table 2, the IC_{50} values of 2,9-bis-substituted-1,10‐phenanthroline derivatives 1a–i were found ranging from 1.3 to 19.0 μ M. Against the human K562 chronic myeloid leukemia cell line, the 2,9-bis[4-(pyridinylalkylaminomethyl)phenyl]-1,10‐phenanthroline derivatives 1a–i showed some significant antiproliferative activities, with IC_{50} determined between 3.9 and 5.3μ M, whereas APTO-253 showed a modest antiproliferative activity of 35.6 μM against this same leukemia cell line. Among the nine 1.10-phenanthrolines 1a-i tested for antiproliferative activities against the basophilic leukemia KU812 cell line, the 1,10‐phenanthroline 1i was found to be the most biologically active compound with an IC_{50} of 1.6 μ M. In

TABLE 1 Physical properties of amines 1a–i

Compound		Salt ^a	MP $(^{\circ}C)^{b}$	% Yield ^c
1a	Pale yellow crystals	2 (COOH) ₂	187-189	75
1 _b	Yellow crystals	2 (COOH) ₂	$222 - 224$	62
1c	Pale yellow crystals	2 (COOH) ₂	180-182	89
1 _d	Yellow crystals	2 (COOH) ₂	182-184	72
1e	Yellow crystals	2(COOH)	181-183	95
1 _f	Yellow crystals	2 (COOH) ₂	187-189	81
1 _g	Yellow crystals	2(COOH)	$141 - 143$	75
1 _h	Yellow crystals	2 (COOH) ₂	134-136	78
1i	Yellow crystals	2(COOH)	139-141	80

^aThe stoichiometry and composition of the salts were determined by elemental analyses, and obtained values were within ±0.4% of the theoretical values.

^bCrystallization solvent: 2-PrOH-H₂O.

^cThe yields only included the conversions into the ammonium oxalates.

TABLE 2 In vitro activity of compounds 1a–i against five human myeloid leukemia cell lines, and cytotoxicity against the HS27a stromal cell line

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Abbreviation: n.d., not done.

addition, the 1,10-phenanthrolines 1g-i, which were disubstituted with a C_3 pyridinyl chain, exhibited a better biological activity than their C_1 or C_2 homologs, compounds $1a-c$ and $1d-f$, respectively, that is, IC_{50} of $1.6-2.5 \mu M$ for $1g-i$ versus 2.4 -2.9 μ M for 1d-f and 2.1-4.6 μ M for 1a-c. Moreover, against the KU812 cell line, derivative 1i bearing pyridin‐4‐ ylpropylaminomethyl side chains at position 4 of the benzyl rings displayed better activity than its analogs substituted with pyridin‐3‐ylpropylaminomethyl or pyridin‐2‐ylpropylaminometh‐ yl side chains (IC₅₀ = 1.6 μ M for 1i versus IC₅₀ = 2.4 and 2.5 μ M for 1h and 1g, respectively). Against the acute myeloid leukemia MV4‐11 cell line, the biological results of compounds 1a–i were

Abbreviations: n.d., not done; SI, selectively index.

^aSI was defined as the ratio between the CC_{50} value against the HS27a normal cells and the IC_{50} value against the leukemia K562, KU812, MV4-11, HL60, or U937 cells.

found quite similar to those noticed against the KU812 cell line in terms of structure–activity relationships. These results exhibited a potent cytotoxicity for disubstituted pyridinylpropylami‐ nomethyl phenanthrolines $1g-i$ (IC₅₀ ranging from 1.3 to 2.1 μ M), greater than that observed for disubstituted pyridinylethylamin‐ omethyl phenanthrolines 1e-f (IC₅₀ = 2.3-2.9 μ M) and for disubstituted pyridinylmethylaminomethyl phenanthrolines 1a–c (IC_{50} = 2.9-3.1 μ M). Surprisingly, disubstituted pyridinylethylaminomethyl phenanthroline 1d showed an IC_{50} of 1.5 μ M in the same range as that observed for pyridinylpropylaminomethyl derivatives 1g–i. Nevertheless, our tested phenanthrolines 1a–i were found less active than the reference APTO‐253 phenanthroline, that is, $IC_{50} = 1.3-3.1 \mu M$ for $1a-i$ versus $0.3 \mu M$ for APTO‐253. Against the HL60 human acute promyeloid leukemia cell line, most of our tested phenanthrolines, 1a–i, showed an antiproliferative activity with IC_{50} values ranging from 3.0 to 8.0 μ M, except 1c that was found inactive. Among derivatives 1a–i, the phenanthrolines derivatives 1g–i, bearing pyridinylpropylaminomethyl moieties, exhibited the best antiproliferative activity against the growth of human myeloid U937 cell line (IC₅₀ ranging from 2 to 3μ M) in comparison with their other analogs, $1a-c$ and $1d-f$, which were found less active (IC₅₀) ranging from 5 to 19 μ M for 1a-c and IC₅₀ ranging from 4 to 7.5μ M for 1d-f), but lower than that of the reference compound APTO-253 (IC₅₀ < 1 μ M). In addition, the cytotoxicity of each phenanthroline, 1a–i, was then evaluated against the HS27a stromal cell line to determine some index of selectivity (Tables 2 and 3). Thus, indexes of selectivity (IS) were defined as the ratio of the IC_{50} value against the HS27a stromal cell line to the IC_{50} value against the K562, KU812, MV4‐11, HL60, and U937 cell lines (Table 3).

As expected, most of the phenanthrolines, 1a–i, showed a significant level of cytotoxicity against the HS27a stromal cell line with IC₅₀ ranging from 3.4 to over 23.6 μ M. Among them, we could notice that 2,9‐bis[4‐(pyridin‐4‐ylmethylaminomethyl)phenyl]‐1,10‐

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TABLE 4 FRET melting values for the selected compounds, 1a-i, and APTO-253 with Fc‐MYCT, FK‐RAST, FBcl‐2T, F21T, and FdxT in K^+ conditions at $2 \mu M$

Note: PhenDC3 was tested at 0.5μ M. Error margins correspond to SD of three replicates.

Abbreviation: FRET, fluorescence resonance energy transfer.

 $\mathrm{^{a} \Delta} T_{\mathrm{m}}$ values of Fc-MYCT, FK-RAST, FBcl-2T, F21T, and FdxT (0.2 μ M) were recorded in 10 mM lithium cacodylate (pH 7.2), 10 mM KCl, 90 mM LiCl.

phenanthroline 1c showed interesting selectivity toward K562, KU812, and MV4-11 cell lines (SI = 6.1, 5.8, and 7.9, respectively). Derivative 1c could now be considered a suitable candidate for further pharmacological studies. The reference compound APTO‐ 253 also showed interesting selectivity with an IS value of 13.3 against the MV4‐11 cell line.

2.3 | FRET melting experiments

G4‐forming motifs have been identified in oncogene promoter regions, bringing to light the therapeutic potential for targeted gene regulation at the transcriptional level. Among them, the most studied oncogene promoter G4s involve c‐MYC, Bcl‐2, h‐Ras, K‐RAS, and c‐KIT. Thus, we evaluated ligand‐induced stabilization, specificity, and selectivity of these newly synthesized phenanthrolines, 1a–i, for various oncogene promoter G4 topologies including c‐MYC, BCL‐2, and K‐RAS through FRET melting experiments (Tables 4 and 5, exemplary melting curves in Figures S1–S4).

The stabilization of our compounds was also investigated against the fluorescently labeled human telomeric sequence F21T. To probe the G4 selectivity of our heterocyclic ligands 1 over duplex DNA, a FRET melting assay was performed using a duplex control sequence, FdxT. For comparison, we also evaluated the reference G4 ligand PhenDC3 and the antileukemial reference drug APTO‐253. To enable comparison of selectivities, we calculated the difference (ΔT_m) between the T_m of the G4 formed by c-MYC, Bcl-2, K-RAS, F21T, or FdxT in the presence or absence of each selected compound. These $\Delta T_{\rm m}$ values are presented in Table 4. For these substituted phenanthroline compounds, $1a-i$, the ΔT_m values ranged from 0.1°C to 17.9°C at 2 µM ligand concentration.

The best nitrogen heterocyclic ligands that stabilize all the three proto-oncogenes (c-MYC, Bcl-2, K-RAS) and the human F21T G4 sequences were compounds 1g-i, bearing pyridinylpropylaminomethyl side chains at position 4 of the benzyl rings (Table 4). We also noticed that 2,9‐bis[4‐(pyridin‐2‐ylmethylaminomethyl)phenyl]‐1,10‐phenanthro‐ line 1a showed the same range of stabilization on the K‐RAS telomeric G4 as those observed for compounds $1g-i$ with a ΔT_m value of 11.7° C versus 8.9–12.4°C for 1g–i. The radar plot shown in Figure 2 shows that compounds 1g–i are able to stabilize most of all the G4‐forming sequences.

As observed in Table 4, the stabilization effects on the different quadruplex sequences seem to be dependent on the length of the pyridinylalkylaminomethyl side chain of the phenanthrolines 1. For each G4 sequence, the substituted phenanthroline ligands 1g-i bearing pyridinylpropylaminomethyl side chains exhibited the best and strong stabilization profile in comparison with their pyridinylethylaminomethyl or pyridinylmethylaminomethyl‐ substituted homologs 1d–f and 1a–c, with the latter being the least stabilizing ligands. Concerning the position of the substitution of the pyridine group on the aminoalkyl chain, we cannot really conclude in terms of structure–activity relationships. In addition, compounds 1a–i have higher affinities toward the K‐RAS human oncogene promoter sequence as compared with c‐MYC and Bcl‐2 ones. From a general point of view, all our ligands, 1a–i, were found to show lower stabilization than the reference PhenDC3 ligand. Surprisingly, by using FRET melting assays, the reference drug APTO‐253 was found nearly ineffective to stabilize all the tested G4 sequences, even if this compound was previously described to stabilize the c-MYC G4 structure.^[16]

In addition, we could also notice that phenanthrolines 1g-i were found more specific for the hybrid G4 topology of the F21T sequence (16.5–17.9°C) than for the parallel G4 structures of c‐MYC, Bcl‐2, or K‐RAS (3.4–7.7°C for c‐MYC; 8.9–12.4°C for K‐RAS, and 6.8–9.2°C for Bcl‐2).

TABLE 5 Labeled DNA sequences used in the present study

Note: All sequences are provided in the 5ʹ to 3ʹ direction.

FIGURE 2 Stabilization specificity profile of $1g-i$ (2 µM) toward various G4 oligonucleotides. The difference in T_m in the presence and absence of $1g-i$, ΔT_m , in °C is plotted for each sequence. Four quadruplexes and one duplex (FdxT) were tested. Experiments were performed in 10 mM lithium cacodylate buffer (pH 7.2), 90 mM LiCl, and 10 mM KCl. T_m in the absence of ligands: FdxT (T_m = 66.6 ± 0.7°C); Fc-mycT (T_m = 64.8 ± 0.3°C); FK-RAST $(T_m = 45.9 \pm 0.0^{\circ}\text{C})$; F21T $(T_m = 52.4 \pm 0.3^{\circ}\text{C})$; FBcl-2T $(T_m = 61.6 \pm 0.5^{\circ}C)$

FRET assays showed that there was no binding to duplex DNA sequence.

2.4 | Detection of telomerase activity in cell lysates

As the human telomeric motif F21T was among the most stabilized G4 sequences (Table 4), we decided to investigate whether our more active phenanthrolines 1g–i may interfere with canonical telomeric functions. For this reason, we assayed their impact on telomerase activity. Telomerase activity in HuT78 and MV4‐11 protein extracts was investigated in the presence of 0–5 µM of the ligands (Figure 3a–c).

As expected at 0 µM, telomerase activity was detectable in both cell lines. The value at 0 µM was arbitrarily considered for 100% telomerase activity. All our three phenanthrolines, 1g–i, practically inhibited telomerase activity at 5 µM for HuT78 and MV4‐11 cell lines. However, at 5 µM, 1i seems to be the more bioactive compound to abolish most of the telomerase activity (>95%) in both leukemic cell lines. Although telomerase activity was detected when tested at 0.5, 1, and $2 \mu M$ of $1g-i$, an 80% decrease was observed at 2μ M for the three tested phenanthrolines, $1g-i$, in both cell lines.

FIGURE 3 Telomerase activity in HuT78 and MV-4-11 cell lysates was investigated in the presence of 0-5 µM of 1g (a), 1h (b), and 1i ligands (c). The p values calculated from Mann–Whitney tests are as follows: ns, not significant, $p \le 0.05$, ** $p > 0.01$

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TABLE 6 DNA oligonucleotides used in native mass spectrometry experiments

Note: All sequences are provided in the 5ʹ to 3ʹ direction.

Abbreviations: dsDNA, double‐stranded DNA; ssDNA, single‐stranded DNA.

^aThe bcl2MidG4 sequence variant has been used.

2.5 | Native electrospray mass spectrometry

Native mass spectrometry experiments were carried out to characterize the affinity and stoichiometry of binding of 1d, 1g, 1h, and 1i to G4forming oligonucleotides, and their selectivity against unfolded and double‐stranded DNA (Figures S1–S5).

The samples were prepared in trimethylammonium acetate (TMAA)‐buffered solutions, supplemented with 1 mM KCl, to allow the formation of G4s with the biologically relevant K^+ cation.^[27] The sequences are similar to those used in the FRET melting experiment, yet the conformers formed in solution might differ owing to the lower potassium concentration and the absence of fluorophore labeling (Table 6). The topologies were, therefore, characterized by circular dichroism in the mass spectrometry conditions (Figures S5–S9). c‐MYC folds into a parallel quadruplex (positive band at 260 nm and negative band at 240 nm) and 24TTG, a 24‐mer variant of the human telomeric sequence, into a hybrid topology (positive band at 290 nm and shoulder at 270 nm). Both adopt the same structure in 1 mM KCl as their PDB‐deposited structures (obtained at higher KCl con $centrations$ ^[31,32] and have a similar profile as the Fc-MYCT and F21T oligonucleotides from the FRET panel. The particular variant of BCI-2 used here (the same as in the FRET panel) can form a mixture of hybrid conformers,^[28] which are almost entirely folded in 1 mM KCl (positive bands at 260 and 290 nm). Finally, we found that the K‐RAS oligonucleotide is almost entirely unfolded in 1 mM KCl solution, as evidenced by the absence of specific potassium cation binding and CD profile (weak band at 260 nm), which was confirmed by UV melting experiments (Figure S10). It was, therefore, used as a G‐rich, unfolded, single‐ stranded control. The selectivity against duplex DNA was assessed by including the auto-complementary ds26 sequence in the panel.

All four ligands bind moderately, and remarkably similar to 24TTG with a 1:1 stoichiometry ($pK_{d1} = 4.6$; Table 7). A weaker second binding site was observed ($pK_{d2} = 3.6-4.0$), with no apparent binding cooperativity $(K_{d1}/K_{d2} = 0.12 - 0.27)$. The same affinity range was found for $1g$ -i binding to BCI-2 (pK $_{d1}$ = 4.5), with the second binding site being very weakly bound (pK $_{d2}$ = 3.2–3.5, K $_{d1}/K_{d2} \le 0.1$). 1d, which has shorter side chains, has a significantly lower affinity $(pK_{d1} = 4.0)$, and no second binding site was observed.

Much larger affinities were found for c-MYC binding by 1h and **1i** (pK_{d1} = 5.5 and 5.0), and to a lower extent by **1g** (pK_{d1} = 4.7). Here again, the shorter side chains of 1d are detrimental to binding ($pK_{d1} = 4.4$). In addition, c-MYC is bound by a second ligand in all cases. These secondary binding sites are weakly bound by 1i and 1g (pK_{d2} = 3.9 and 3.5; K_{d1}/K_{d2} < 0.1), and they remain virtually unbound by 1d (p K_{d2} = 3.0; K_{d1}/K_{d2} = 0.04). Similar to the first binding site, 1h binds better than its counterparts to this secondary site, by an order of magnitude; however, this remains a non‐cooperative process $(pK_{d2} = 4.6; K_{d1}/K_{d2} = 0.13).$

The unfolded form of K-RAS (i.e., not binding K^+) is unbound by all ligands, indicating that they do not bind unfolded ssDNA. However, ligand 1h binds a small amount of this sequence folded into a quadruplex (i.e., binding specifically 2K⁺), which was present in the solution initially and/or by promoting its formation. However, as the amount of formed G4 is not known, this precludes determining the affinity constant for the folded KRAS G4. The apparent pK_d

TABLE 7 Binding affinities of the 1:1 (pK_{d1}) and 2:1 (pK_{d2}) complexes, and binding cooperativities (K_{d1}/K_{d2})^a, of 1d and 1g-i ligands with 24TTG, BCI-2, c-MYC, K-RAS in 100 mM TMAA and 1 mM KCI, determined by native MS

	24TTG				BCI-2			c-MYC			ds ₂₆					
Ligand	1 _d	1 _g	1h	1i	1 _d	1g	1h	11	1 _d	1g	1h	1i	1 _d	1 _g	1 _h	1i
pK_{d1}	4.6	4.6	4.6	4.6	4.0	4.5	4.5	4.5	4.4	4.7	5.5	5.0	n.b.	n.b.	n.b.	n.b.
pK_{d2}	3.9	4.0	4.0	3.6	n.b	3.2	3.5	3.3	3.0	3.5	4.6	3.9	n.b.	n.b.	n.b.	n.b.
K_{d1}/K_{d2}	0.20	0.27	0.27	0.12	n.a	0.05	0.10	0.07	0.04	0.07	0.13	0.08	n.a.	n.a.	n.a.	n.a.

Abbreviations: MS, mass spectrometry; TMAA, trimethylammonium acetate.

^an.a., not applicable; n.b., no observed binding.

FIGURE 4 Comparative docking of ligands 1h and 1b on the c-myc G-quadruplex extracted from PDB structure 2MGN^[33]

obtained is the convolution of the folding equilibrium (displaced mainly toward the unfolded form) and the binding equilibrium. For this reason, it cannot be compared directly with the pK_d values obtained for the three other G4s. Finally, no complex was observed on ds26, confirming that this set of ligands does not bind to dsDNA.

Overall, ligands 1g–i share similar binding behaviors, that is, (i) no binding to ss‐ and ds‐DNA, (ii) moderate binding to the hybrid quadruplexes formed by 24TTG and BCl‐2, (iii) stronger binding to the parallel c‐MYC quadruplex, and (iv) weak binding to secondary binding sites. Ligand 1d, which has shorter side chains, has worse affinities for BCl‐2 and c‐MYC than its counterparts, in line with the FRET melting results. This could be explained by a lesser ability to form electrostatic interactions with loop or groove phosphates for some topologies, but not all (24TTG is a counterexample). Ligand 1h displays the strongest binding for all tested quadruplexes and is capable of folding small amounts of, and/or promoting the folding of, the K‐RAS G4.

2.6 | Docking

Ligands with the longer arms (1g–i) have the highest affinity for the G4s. To visualize where this arm length places the pyridinyl groups, we carried out docking with a parallel c‐myc G4. The structure of the complex between the oligonucleotide dTGAGGGTGGTGAGGGTGGGGAAGG and PhenDC3 (PDB code: $2MGN$ ^[33] was used as the starting point. Ligands 1h (long arms) and 1b (short arms control) were constructed and optimized using the AMBER ParmBSC0 force field.^[34] Three different conformers of 1h and 1b were tested and energy-ranked (Figures S11 and S12). The PhenDC3 ligand on the 2MGN structure was then replaced with 1h or 1b, and the geometry was further optimized until a gradient of 0.05 kcal/mol was reached, and then it was subjected to 2 ns molecular dynamics. HyperChem 8.0.10 (Hypercube Inc.) was used for the calculations. The figures were generated using Biovia Discovery Studio Visualizer (Dassault Systèmes). Figure 4, showing the lowest energy conformers of 1h and 1b stacked on the terminal quartet, reveals that in the most affine ligand (1h), the pyridinyl groups dangle in the groove.

3 | CONCLUSION

In this study, we have designed and synthesized a series of nine new 2,9‐bis[4‐(pyridinylalkylaminomethyl)phenyl]‐1,10‐phenanthroline derivatives, 1a–i, and evaluated their antileukemic activity against five human leukemic cell lines, K562, KU812, MV4‐11, HL60, and U937. Their cytotoxicity against the HS27a stromal cell line was also evaluated. Their ability to stabilize various oncogene promoter G4 structures (c‐MYC, BCL‐2, and k‐RAS) has been also determined through the FRET melting assay and native mass spectrometry. By using these pharmacological and biophysical results, preliminary SAR studies on these phenanthroline compounds have been discussed. The biological evaluations of new derivatives 1a–i showed cytotoxic activity against the different myeloid and lymphoid leukemia cell lines. The disubstituted pyridinylpropylaminomethyl phenanthrolines 1g–i showed the more interesting antileukemic activities and may be now involved in further pharmacological modulations and investigations. The inhibition for telomerase activity by the more active ligands was assessed through telomeric repeat amplification protocol (TRAP) assay, and the result showed a significant inhibition at 5 µM concentration for ligand 1i. Moreover, further investigations are required to reveal the complete mechanism of action of these bioactive phenanthrolines.

EXPERIMENTAL

4.1 | Chemistry

4.1.1 | General

Commercially available reagents were used as received without additional purification. The reference drug APTO‐253 (free base) was provided from CliniSciences. Melting points were determined with an SM-LUX-POL Leitz hot-stage microscope and were uncorrected. Infrared (IR) spectra were recorded on a NICOLET 380FT‐IR spectrophotometer. Nuclear magnetic resonance (NMR) spectra were recorded with tetramethylsilane as an internal standard using a Bruker Avance 300 spectrometer. Splitting patterns have been reported as follows: s, singlet; bs, broad singlet; d, doublet; t, triplet; q, quartet; dd, double doublet; ddd, double double doublet; qt, quintuplet; m, multiplet. Analytical thin‐layer chromatography was carried out on 0.25 mm precoated silica gel plates (POLYGRAM SIL G/UV254) and visualization of compounds was done after UV light irradiation. Silica gel 60 (70–230 mesh) was used for column chromatography. Mass spectra were recorded on an ESI LTQ Orbitrap Velos mass spectrometer (Thermo Fisher Scientific). Ionization was performed using an electrospray ion source operating in a positive ion mode with a capillary voltage of 3.80 kV and capillary temperature of 250°C. The scan type analyzed was full scan, and all mass spectrometry (MS) recordings were in the m/z range between 150 and 2000 m/z. Fragmentation of no type was performed, and the resolution used for the analysis was 60.000.

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The InChI codes of the investigated compounds, together with some biological activity data, are provided as Supporting Information.

4.1.2 | General procedure for the synthesis of 2,9‐bis[4‐(pyridinylalkyliminomethyl)phenyl]‐1,10‐ phenanthrolines 4a–i

The bis‐[2,9‐(formylphenyl)]‐1,10‐phenanthroline 3 (150 mg, 0.386 mmol) was dissolved in 6 ml of toluene. Activated molecular sieves 4 Å (800 mg) and then pyridinylalkylamine (0.812 mmol) were introduced in the latter solution. The reaction mixture was stirred in a stoppered flask for 24 h. The suspension that was obtained was filtered and washed with dichloromethane. Then, the solvent was removed under reduced pressure to afford diimine 4 as an oil.

2,9‐Bis[4‐(pyridin‐2‐ylmethyliminomethyl)phenyl]‐1,10‐ phenanthroline (4a)

Yellow oil (yield, 98%). 1 H NMR (CDCl₃) δ ppm: 8.62 (m, 4H, 2H-6_{pyr} and 2CH═N), 8.57 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.35 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.21 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 8.06 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.82 (s, 2H, H‐5 and H‐6), 7.73 (ddd, 2H, J = 7.80, 7.20 and 1.80 Hz, 2H-4_{pyr}), 7.52 (ddd, 2H, J = 7.20, 0.90 and 0.90 Hz, 2H-3_{pyr}), 7.22 (ddd, 2H, J = 7.80, 6.00 and 0.90 Hz, $2H-5_{\text{pvr}}$, 5.05 (s, 4H, 2NCH₂).

2,9‐Bis[4‐(pyridin‐3‐ylmethyliminomethyl)phenyl]‐1,10‐ phenanthroline (4b)

Yellow crystals (yield, 98%); m.p. = 195-197°C. 1 H NMR (CDCl₃) δ ppm: 8.67 (ddd, 2H, J = 1.90, 0.80 and 0.80 Hz, 2H-2_{pyr}), 8.56-8.53 (m, 8H, 2H-2′, 2H-6′, 2H-6_{pyr} and 2CH=N), 8.29 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.16 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 8.00 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.77 (s, 2H, H‐5 and H‐6), 7.73 (ddd, 2H, J = 7.65, 1.90 and 1.90 Hz, 2H-4_{pyr}), 7.31 (ddd, 2H, J = 7.65, 5.40 and 0.80 Hz, 2H-5_{pyr}), 4.88 (s, 4H, 2NCH₂).

2,9‐Bis[4‐(pyridin‐4‐ylmethyliminomethyl)phenyl]‐1,10‐ phenanthroline (4c)

Beige crystals (yield, 98%); m.p. = 235-237°C. ¹H NMR (CDCl₃) δ ppm: 8.62 (dd, 4H, J = 6.00 and 1.50 Hz, 2H-2_{pyr} and 2H-6_{pyr}), 8.58 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.55 (s, 2H, 2CH═N), 8.36 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.21 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 8.04 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.84 (s, 2H, H‐5 and H‐6), 7.37 (dd, 4H, J = 6.00 and 1.50 Hz, 2H-3_{pyr} and 2H-5_{pyr}), 4.89 (s, 4H, $2NCH₂$).

2,9‐Bis[4‐(pyridin‐2‐ylethyliminomethyl)phenyl]‐1,10‐ phenanthroline (4d)

Yellow crystals (yield, 98%); m.p. = $166-168$ °C. 1 H NMR (CDCI₃) δ ppm: 8.59 (ddd, 2H, J = 6.00, 1.80 and 0.90 Hz, 2H-6_{pyr}), 8.50 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.31 (s, 2H, 2CH═N), 8.27 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.13 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.90 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.75 (s, 2H, H‐5 and H‐6), 7.58 (ddd, 2H, J = 7.80, 7.20 and 1.80 Hz, 2H-4_{pyr}), 7.22 (ddd, 2H, J = 7.20, 0.90 and 0.90 Hz, 2H-3_{pyr}), 7.11 (ddd, 2H, J = 7.80, 6.00 and 0.90 Hz, 2H-5_{pyr}), 4.09 (t, 4H, J = 7.20 Hz, 2NCH₂), 3.26 (t, 4H, J = 7.20 Hz, $2CH₂Pyr$).

2,9‐Bis[4‐(pyridin‐3‐ylethyliminomethyl)phenyl]‐1,10‐ phenanthroline (4e)

Pale-yellow oil (yield, 98%). ¹H NMR (CDCl₃) δ ppm: 8.57 (ddd, 2H, J = 1.90, 0.90 and 0.90 Hz, 2H-2_{pyr}), 8.53 (d, 2H, J = 8.40 Hz, 2H-2['] and 2H-6 $\frac{1}{2}$, 8.49 (dd, 2H, J = 5.90 and 1.50 Hz, 2H-6_{pyr}), 8.33 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.29 (s, 2H, 2CH═N), 8.18 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.93 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.81 (s, 2H, H‐5 and H‐6), 7.59 (ddd, 2H, J = 7.80, 1.80 and 1.80 Hz, 2H-4_{pyr}), 7.24 (ddd, 2H, J = 7.80, 5.90 and 0.90 Hz, 2H-5_{pyr}), 3.94 (t, 4H, $J = 7.05$ Hz, 2NCH₂), 3.10 (t, 4H, $J = 7.05$ Hz, 2CH₂Pyr).

2,9‐Bis[4‐(pyridin‐4‐ylethyliminomethyl)phenyl]‐1,10‐ phenanthroline (4f)

Pale-yellow crystals (yield, 97%); m.p. = $160-162^{\circ}$ C. ¹H NMR (CDCl₃) δ ppm: 8.56 (dd, 4H, J = 6.00 and 1.50 Hz, 2H-2_{pyr} and 2H-6_{pyr}), 8.53 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.36 (d, 2H, J = 8.55 Hz, H‐4 and H‐7), 8.29 (s, 2H, 2CH═N), 8.21 (d, 2H, J = 8.55 Hz, H‐3 and H‐8), 7.94 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.84 (s, 2H, H‐5 and H‐6), 7.23 (dd, 4H, J = 6.00 and 1.50 Hz, 2H-3_{pyr} and 2H-5_{pyr}), 3.96 (t, 4H, J = 6.90 Hz, $2NCH_2$), 3.10 (t, 4H, J = 6.90 Hz, 2CH₂Pyr).

2,9‐Bis[4‐(pyridin‐2‐ylpropyliminomethyl)phenyl]‐1,10‐ phenanthroline (4g)

Yellow crystals (yield, 93%); m.p. = 153-155°C. ¹H NMR (CDCl₃) δ ppm: 8.52 (ddd, 2H, J = 6.10, 1.80 and 0.90 Hz, 2H-6_{pyr}), 8.49 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.35 (s, 2H, 2CH═N), 8.20 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.08 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.92 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.67 (s, 2H, H‐5 and H‐6), 7.55 (ddd, 2H, J = 7.80, 7.20 and 1.80 Hz, 2H-4_{pyr}), 7.16 (ddd, 2H, J = 7.20, 0.90 and 0.90 Hz, 2H-3_{pyr}), 7.06 (ddd, 2H, J = 7.80, 6.10 and 0.90 Hz, 2H-5_{pyr}), 3.71 (t, 4H, J = 6.90 Hz, 2NCH₂), 2.92 (t, 4H, J = 6.90 Hz, 2CH₂Pyr), 2.19 (qt, 4H, $J = 6.90$ Hz, $2CH_2$).

2,9‐Bis[4‐(pyridin‐3‐ylpropyliminomethyl)phenyl]‐1,10‐ phenanthroline (4h)

Yellow oil (yield, 98%). ¹Η NMR (CDCl₃) δ ppm: 8.55-8.51 (m, 6H, 2H- 2_{pvr} , 2H-2' and 2H-6'), 8.45 (dd, 2H, J = 5.70 and 1.20 Hz, 2H-6_{pvr}), 8.34 (s, 2H, 2CH=N), 8.18 (d, 2H, J = 8.40 Hz, H-4 and H-7), 8.07 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.94 (d, 4H, J = 8.10 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.66 (s, 2H, H-5 and H-6), 7.51 (ddd, 2H, J = 7.70, 1.80 and 1.80 Hz, 2H-4_{pyr}), 7.25–7.20 (m, 2H, 2H-5_{pyr}), 3.66 (t, 4H, J = 7.10 Hz, 2NCH₂), 2.73 (t, 4H, $J = 7.10$ Hz, 2CH₂Pyr), 2.07 (qt, 4H, $J = 7.10$ Hz, 2CH₂).

2,9‐Bis[4‐(pyridin‐4‐ylpropyliminomethyl)phenyl]‐1,10‐ phenanthroline (4i)

Yellow oil (yield, 98%). ¹H NMR (CDCl₃) δ ppm: 8.53 (dd, 4H, J = 6.00 and 1.50 Hz, 2H-2_{pyr} and 2H-6_{pyr}), 8.49 (d, 4H, J = 8.40 Hz, 2H-2['] and 2H‐6ʹ), 8.34 (s, 2H, 2CH═N), 8.26 (d, 2H, J = 8.55 Hz, H‐4 and H‐7), 8.13 (d, 2H, J = 8.55 Hz, H‐3 and H‐8), 7.93 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H-5′), 7.73 (s, 2H, H-5 and H-6), 7.14 (dd, 4H, J = 6.00 and 1.50 Hz, 2H-3_{pyr} and 2H-5_{pyr}), 3.67 (t, 4H, J = 6.90 Hz, 2NCH₂), 2.73 (t, 4H, J = 6.90 Hz, 2CH₂Pyr), 2.08 (qt, 4H, J = 6.90 Hz, 2CH₂).

4.1.3 | General procedure for the synthesis of 2,9bis[4‐(pyridinylalkylaminomethyl)phenyl]‐1,10‐ phenanthrolines 1a–i

To a solution of compounds 4a–i (0.4 mmol) in methanol (10 ml), sodium borohydride (2.4 mmol, 6 eq.) was added portion‐wise at 0°C. The reaction mixture was then stirred at room temperature for 2 h. Then it was evaporated to dryness under reduced pressure. After cooling, the residue was triturated in water and extracted with dichloromethane (40 ml). The organic layer was separated, dried over sodium sulfate and activated charcoal, and evaporated to dryness. The residue was then purified by column chromatography on silica gel using dichloromethane/methanol (90:10 v/v) as an eluent to give the pure products 1a–i.

2,9‐Bis[4‐(pyridin‐2‐ylmethylaminomethyl)phenyl]‐1,10‐ phenanthroline (1a)

Pale-yellow crystals (yield, 90%); m.p. = 69-71°C. 1 H NMR (CDCl₃) δ ppm: 8.60 (ddd, 2H, J = 5.90, 1.80 and 0.90 Hz, 2H-6_{pyr}), 8.45 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.29 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.14 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.77 (s, 2H, H‐5 and H‐6), 7.66 (ddd, 2H, J = 7.80, 7.20 and 1.50 Hz, 2H-4_{pvr}), 7.60 (d, 4H, J = 8.40 Hz, 2H-3['] and 2H-5¹), 7.36 (ddd, 2H, J = 7.20, 0.90 and 0.90 Hz, 2H-3_{pyr}), 7.18 (ddd, 2H, J = 7.80, 5.90 and 0.90 Hz, 2H-5_{pyr}), 4.00 (s, 4H, 2NCH₂), 3.98 (s, 4H, 2NCH₂); ¹³C NMR (CDCl₃) δ ppm: 161.3 (C-2_{pyr}), 158.0 (C-2), 150.7 (C-6pyr), 147.5 (C‐1a and C‐10a), 142.9 (C‐1ʹ and C‐4ʹ), 139.6 (C‐4a and C‐ 6a), 138.2 (C-4 and C-7), 137.9 (C-4_{pyr}), 130.1 (C-3' and C-5'), 129.1 (C-2' and C-6′), 127.3 (C-5 and C-6), 123.8 (C-3_{pyr}), 123.4 (C-5_{pyr}), 121.2 (C-3 and C-8), 55.9 (NCH₂), 54.6 (NCH₂); matrix-assisted laser desorption/ ionization-time of flight (MALDI-TOF) MS m/z [M+H]⁺ calcd. for C38H33N6: 573.276, found: 573.276.

2,9‐Bis[4‐(pyridin‐3‐ylmethylaminomethyl)phenyl]‐1,10‐ phenanthroline (1b)

Yellow oil (yield, 97%). ¹H NMR (CDCl₃) δ ppm: 8.60 (ddd, 2H, J = 1.80, 0.90 and 0.90 Hz, 2H-2_{pyr}), 8.53 (dd, 2H, J = 6.00 and 1.50 Hz, 2H-6_{pyr}), 8.46 (d, 4H, J = 8.40 Hz, 2H-2' and 2H-6'), 8.31 (d, 2H, J = 8.40 Hz, H-4 and H‐7), 8.15 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.79 (s, 2H, H‐5 and H‐6), 7.74 (ddd, 2H, J = 7.80, 1.80 and 1.80 Hz, 2H-4_{pyr}), 7.57 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.29 (ddd, 2H, J = 7.80, 6.00 and 0.90 Hz, 2H-5_{pyr}), 3.94 (s, 4H, 2NCH₂), 3.87 (s, 4H, 2NCH₂), 1.86 (bs, 2H, 2NH); ¹³C NMR (CDCl₃) δ ppm: 157.9 (C-2), 151.2 (C-2_{pyr}), 149.9 (C-6_{pyr}), 147.5 (C‐1a and C‐10a), 142.7 (C‐1ʹ and C‐4ʹ), 139.8 (C‐3pyr), 138.3 (C‐4 and C‐ 7), 137.3 (C-4_{pyr}), 137.0 (C-4a and C-6a), 130.0 (C-3['] and C-5[']), 129.2 (C-2' and C-6', 127.4 (C-5 and C-6), 124.8 (C-5_{pvr}), 121.3 (C-3 and C-8), 54.3 (NCH₂), 51.7 (NCH₂); MALDI-TOF MS m/z [M+H]⁺ calcd. for C₃₈H₃₃N₆: 573.276, found: 573.276.

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2,9‐Bis[4‐(pyridin‐4‐ylmethylaminomethyl)phenyl]‐1,10‐ phenanthroline (1c)

Yellow oil (yield, 83%). ¹H NMR (CDCl₃) δ ppm: 8.53 (dd, 4H, J = 6.30 and 1.80 Hz, 2H-2_{pyr} and 2H-6_{pyr}), 8.42 (d, 4H, J = 8.40 Hz, 2H-2['] and 2H‐6ʹ), 8.25 (d, 2H, J = 8.55 Hz, H‐4 and H‐7), 8.09 (d, 2H, J = 8.55 Hz, H‐3 and H‐8), 7.73 (s, 2H, H‐5 and H‐6), 7.52 (d, 4H, J = 8.40 Hz, 2H‐ 3' and 2H-5 $\frac{1}{2}$, 7.28 (dd, 4H, J = 6.30 and 1.80 Hz, 2H-3_{pyr} and 2H- 5_{pyr}), 3.87 (s, 4H, 2NCH₂), 3.80 (s, 4H, 2NCH₂), 1.65 (bs, 2H, 2NH); ¹³C NMR (CDCl₃) δ ppm: 157.8 (C-2), 151.1 (C-2_{pyr} and C-6_{pyr}), 150.8 (C-4_{pyr}), 147.4 (C-1a and C-10a), 142.6 (C-1['] and C-4[']), 139.8 (C‐4a and C‐6a), 138.3 (C‐4 and C‐7), 130.0 (C‐3ʹ and C‐5ʹ), 129.1 (C‐ 2' and C-6'), 127.4 (C-5 and C-6), 124.4 (C-3_{pyr} and C-5_{pyr}), 121.2 (C-3 and C-8), 54.2 (NCH₂), 53.0 (NCH₂); MALDI-TOF MS m/z [M+H]⁺ calcd. for $C_{38}H_{33}N_6$: 573.276, found: 573.276.

2,9‐Bis[4‐(pyridin‐2‐ylethylaminomethyl)phenyl]‐1,10‐ phenanthroline (1d)

Pale-yellow oil (yield, 67%). 1 H NMR (CDCl₃) δ ppm: 8.55 (ddd, 2H, J = 5.90, 2.10 and 1.20 Hz, 2H-6_{pyr}), 8.42 (d, 4H, J = 8.25 Hz, 2H-2['] and 2H-6′), 8.26 (d, 2H, J = 8.40 Hz, H-4 and H-7), 8.11 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.74 (s, 2H, H‐5 and H‐6), 7.59 (ddd, 2H, J = 7.80, 7.20 and 2.10 Hz, 2H-4_{pyr}), 7.53 (d, 4H, J = 8.25 Hz, 2H-3['] and 2H-5 $\frac{1}{2}$, 7.19 (ddd, 2H, J = 7.20, 1.20 and 1.20 Hz, 2H-3_{pyr}), 7.11 (ddd, 2H, J = 7.80, 5.90 and 1.20 Hz, 2H-5_{pyr}), 3.96 (s, 4H, 2NCH₂), 3.11 (t, 4H, J = 6.70 Hz, 2NCH₂), 3.06 (t, 4H, J = 6.70 Hz, 2CH₂Pyr), 1.94 (bs, 2H, 2NH); ¹³C NMR (CDCl₃) δ ppm: 161.6 (C-2_{pyr}), 158.0 (C-2), 150.7 (C-6_{pvr}), 147.5 (C-1a and C-10a), 143.2 (C-1['] and C-4[']), 139.5 (C-4a and C-6a), 138.2 (C-4 and C-7), 137.7 (C-4_{pvr}), 129.9 (C-3' and C-5'), 129.1 (C-2' and C-6'), 127.3 (C-5 and C-6), 124.7 (C-3_{pyr}), 123.6 (C-5_{pyr}), 121.2 (C-3 and C-8), 55.0 (NCH₂), 50.2 (NCH₂), 39.9 (CH₂Pyr); MALDI-TOF MS m/z [M+H]⁺ calcd. for C₄₀H₃₇N₆: 601.308, found: 601.307.

2,9‐Bis[4‐(pyridin‐3‐ylethylaminomethyl)phenyl]‐1,10‐ phenanthroline (1e)

Yellow oil (yield, 61%). ¹H NMR (CDCl₃) δ ppm: 8.48-8.38 (m, 8H, 2H-2_{pyr}, 2H-6_{pyr}, 2H-2['] and 2H-6[']), 8.23 (d, 2H, J = 8.40 Hz, H-4 and H‐7), 8.07 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.69 (s, 2H, H‐5 and H‐6), 7.52–7.43 (m, 6H, 2H-4_{pyr}, 2H-3' and 2H-5'), 7.18 (ddd, 2H, J = 7.80, 6.00 and 0.90 Hz, 2H-5_{pyr}), 3.87 (s, 4H, 2NCH₂), 2.88 (t, 4H, $J = 6.90$ Hz, 2NCH₂), 2.78 (t, 4H, $J = 6.90$ Hz, 2CH₂Pyr), 1.75 (bs, 2H, 2NH); ¹³C NMR (CDCl₃) δ ppm: 157.8 (C-2), 151.5 (C-2_{pyr}), 149.0 (C- 6_{pvr} , 147.4 (C-1a and C-10a), 142.9 (C-1['] and C-4[']), 139.7 (C-3_{pvr}), 138.3 (C‐4 and C‐7), 137.6 (C‐4pyr), 136.9 (C‐4a and C‐6a), 129.9 (C‐ 3' and C-5'), 129.1 (C-2' and C-6'), 127.4 (C-5 and C-6), 124.8 (C-5_{pyr}), 121.2 (C-3 and C-8), 54.9 (NCH₂), 51.4 (NCH₂), 34.9 (CH₂Pyr); MALDI-TOF MS m/z [M+H]⁺ calcd. for C₄₀H₃₇N₆: 601.308, found: 601.308.

2,9‐Bis[4‐(pyridin‐4‐ylethylaminomethyl)phenyl]‐1,10‐

phenanthroline (1f)

Yellow oil (yield, 97%). ¹H NMR (CDCl₃) δ ppm: 8.49 (dd, 4H, J = 6.15 and 1.65 Hz, 2H-2_{pyr} and 2H-6_{pyr}), 8.41 (d, 4H, J = 8.10 Hz, 2H-2['] and

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2H‐6ʹ), 8.25 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.10 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.73 (s, 2H, H‐5 and H‐6), 7.48 (d, 4H, J = 8.10 Hz, 2H‐ $3'$ and 2H-5′), 7.12 (dd, 4H, J = 6.15 and 1.65 Hz, 2H-3_{pyr} and 2H- 5_{pyr}), 3.89 (s, 4H, 2NCH₂), 2.92 (t, 4H, J = 6.60 Hz, 2NCH₂), 2.81 (t, 4H, J = 6.60 Hz, 2CH₂Pyr), 1.91 (bs, 2H, 2NH); ¹³C NMR (CDCl₃) δ ppm: 157.9 (C-2), 151.2 (C-2_{pyr} and C-6_{pyr}), 150.5 (C-4_{pyr}), 147.4 (C-1a and C‐10a), 142.8 (C‐1ʹ and C‐4ʹ), 139.7 (C‐4a and C‐6a), 138.3 (C‐ 4 and C‐7), 129.9 (C‐3ʹ and C‐5ʹ), 129.1 (C‐2ʹ and C‐6ʹ), 127.3 (C‐5 and C-6), 125.6 (C-3_{pyr} and C-5_{pyr}), 121.2 (C-3 and C-8), 54.9 (NCH₂), 50.6 (NCH₂), 37.2 (CH₂Pyr); MALDI-TOF MS m/z [M+H]⁺ calcd. for $C_{40}H_{37}N_6$: 601.308, found: 601.308.

2,9‐Bis[4‐(pyridin‐2‐ylpropylaminomethyl)phenyl]‐1,10‐ phenanthroline (1g)

Yellow oil (yield, 97%). 1 H NMR (CDCl₃) δ ppm: 8.48 (ddd, 2H, $J = 6.00$, 1.80 and 0.90 Hz, 2H-6_{pyr}), 8.39 (d, 4H, $J = 8.40$ Hz, 2H-2['] and 2H-6′), 8.19 (d, 2H, J = 8.70 Hz, H-4 and H-7), 8.05 (d, 2H, J = 8.70 Hz, H‐3 and H‐8), 7.66 (s, 2H, H‐5 and H‐6), 7.53 (ddd, 2H, J = 7.80, 7.20 and 1.80 Hz, 2H-4_{pyr}), 7.50 (d, 4H, J = 8.40 Hz, 2H-3['] and 2H-5 $\frac{1}{2}$, 7.09 (ddd, 2H, J = 7.20, 0.90 and 0.90 Hz, 2H-3_{pyr}), 7.03 (ddd, 2H, J = 7.80, 6.00 and 0.90 Hz, 2H-5_{pyr}), 3.86 (s, 4H, 2NCH₂), 2.84 (t, 4H, J = 7.20 Hz, 2NCH₂), 2.71 (t, 4H, J = 7.20 Hz, 2CH₂Pyr), 2.07 (bs, 2H, 2NH), 1.96 (qt, 4H, J = 7.20 Hz, 2CH₂); ¹³C NMR (CDCl₃) δ ppm: 163.2 (C-2_{pvr}), 157.9 (C-2), 150.5 (C-6_{pvr}), 147.4 (C-1a and C‐10a), 143.2 (C‐1ʹ and C‐4ʹ), 139.5 (C‐4a and C‐6a), 138.2 (C‐ 4 and C-7), 137.7 (C-4_{pyr}), 129.9 (C-3' and C-5'), 129.0 (C-2' and C-6'), 127.3 (C-5 and C-6), 124.2 (C-3_{pyr}), 122.4 (C-5_{pyr}), 121.2 (C-3 and C-8), 55.0 (NCH₂), 50.1 (NCH₂), 37.4 (CH₂Pyr), 31.4 (CH₂); MALDI-TOF MS m/z [M+H]⁺ calcd. for $C_{42}H_{41}N_6$: 629.339, found: 629.506.

2,9‐Bis[4‐(pyridin‐3‐ylpropylaminomethyl)phenyl]‐1,10‐ phenanthroline (1h)

Yellow oil (yield, 86%). 1 H NMR (CDCl₃) δ ppm: 8.46 (ddd, 2H, J = 1.80, 0.90 and 0.90 Hz, 2H-2_{pyr}), 8.44-8.40 (m, 6H, 2H-6_{pyr}, 2H-2ʹ and 2H‐6ʹ), 8.26 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.11 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.74 (s, 2H, H‐5 and H‐6), 7.52 (d, 4H, J = 8.40 Hz, 2H‐3ʹ and 2H‐5ʹ), 7.47 (ddd, 2H, J = 7.80, 1.80 and 1.80 Hz, 2H-4_{pyr}), 7.18 (ddd, 2H, J = 7.80, 5.70 and 0.90 Hz, 2H- 5_{pyr}), 3.88 (s, 4H, 2NCH₂), 2.70 (t, 4H, J = 7.05 Hz, 2NCH₂), 2.68 (t, 4H, J = 7.05 Hz, 2CH₂Pyr), 1.91 (bs, 2H, 2NH), 1.85 (qt, 4H, J = 7.05 Hz, 2CH₂); ¹³C NMR (CDCl₃) δ ppm: 157.8 (C-2), 151.2 (C- 2_{pyr} , 148.6 (C-6_{pyr}), 147.4 (C-1a and C-10a), 143.1 (C-1['] and C-4[']), 139.5 (C‐3pyr), 138.7 (C‐4a and C‐6a), 138.2 (C‐4 and C‐7), 137.1 (C‐4pyr), 129.9 (C‐3ʹ and C‐5ʹ), 129.1 (C‐2ʹ and C‐6ʹ), 127.2 (C‐5 and C-6), 124.6 (C-5_{pyr}), 121.1 (C-3 and C-8), 55.0 (NCH₂), 49.8 (NCH₂), 32.7 (CH₂Pyr), 32.0 (CH₂); MALDI-TOF MS m/z [M+H]⁺ calcd. for C42H41N6: 629.339, found: 629.423.

2,9‐Bis[4‐(pyridin‐4‐ylpropylaminomethyl)phenyl]‐1,10‐ phenanthroline (1i)

Pale-yellow oil (yield, 98%). 1 H NMR (CDCl₃) δ ppm: 8.47 (dd, 4H, J = 6.10 and 1.50 Hz, 2H-2_{pyr} and 2H-6_{pyr}), 8.42 (d, 4H, J = 8.40 Hz, 2H‐2ʹ and 2H‐6ʹ), 8.27 (d, 2H, J = 8.40 Hz, H‐4 and H‐7), 8.11 (d, 2H, J = 8.40 Hz, H‐3 and H‐8), 7.73 (s, 2H, H‐5 and H‐6), 7.51 (d, 4H, J = 8.40 Hz, 2H-3' and 2H-5'), 7.09 (dd, 4H, J = 6.10 and 1.50 Hz, 2H‐3pyr and 2H‐5pyr), 3.87 (s, 4H, 2NCH2), 2.67 (t, 4H, $J = 7.10$ Hz, 2NCH₂), 2.65 (t, 4H, $J = 7.10$ Hz, 2CH₂Pyr), 1.83 (qt, 4H, J = 7.10 Hz, 2CH₂), 1.65 (bs, 2H, 2NH); ¹³C NMR (CDCl₃) δ ppm: 157.9 (C-2), 152.3 (C-4_{pyr}), 151.1 (C-2_{pyr} and C-6_{pyr}), 147.4 (C‐1a and C‐10a), 143.2 (C‐1ʹ and C‐4ʹ), 139.6 (C‐4a and C‐6a), 138.3 (C-4 and C-7), 129.9 (C-3' and C-5'), 129.1 (C-2' and C-6'), 127.3 (C-5 and C-6), 125.3 (C-3_{pyr} and C-5_{pyr}), 121.2 (C-3 and C-8), 55.1 (NCH₂), 49.9 (NCH₂), 34.3 (CH₂Pyr), 32.0 (CH₂); MALDI-TOF MS m/z [M+H]⁺ calcd. for $C_{42}H_{41}N_6$: 629.339, found: 629.339.

4.2 | Biological evaluation

4.2.1 | Cell cultures and reagents

KU812, K562, and MV4‐11 cell lines were obtained from the Deutshe Sammlung von Mikroorganismens und Zellkulturen (DSMZ). HL60, U937, HS‐27, and HuT78 cell lines were obtained from the American Type Culture Collection (ATCC).

BM‐mesenchymal stem cells (MSCs) from brain dead donors were isolated and cultured, as described. Informed consent was obtained before BM samples were taken. MSCs were cultured in αMEM with 2 mM glutamine, 100 U/ml penicillin, 100 µg/ml streptomycin, 0.005% fibroblast growth factor 2 (Gibco–Thermo Fisher Scientific), and 10% fetal bovine serum (Gibco–Thermo Fisher Scientific), at 37°C, 5% CO_{2.}

4.2.2 | Cell proliferation assays

Cell viability and proliferation were studied using either MTS (3‐(4,5‐ dimethylthiazol‐2‐yl)‐5‐(3‐carboxymethoxyphenyl)‐2‐(4‐sulfophenyl)‐2H‐ tetrazolium) or MTT (3‐(4,5‐dimethylthiazol‐2‐yl)‐2,5‐diphenyltetrazolium bromide) cell proliferation assays.

The MTS cell proliferation assay (Promega) is a colorimetric assay system that measures the reduction of a tetrazolium component (MTS) into formazan produced by the mitochondria of viable cells. Cells were washed twice in phosphate‐buffered saline (PBS) and plated in quadruplicate into microtiter plate wells in 100 μ l (10⁴ leukemic cells) of culture media with or without our various compounds at increasing concentrations (0, 1, 5, 10, 20, and 50 μ M) for 1, 2, and 3 days. After 3 h of incubation at 37°C with 20 μl MTS/well, the plates were read by using an ELISA microplate reader (iMark; Bio‐Rad) at 490 nm wavelength. The amount of color produced was directly proportional to the number of viable cells. The results are expressed as the concentrations inhibiting cell growth by 50% after a 3‐day incubation period. The 50% inhibitory concentrations (IC $_{50}$) were determined by linear regression analysis, expressed in μ M \pm SD (Microsoft Excel).

For MTT cell proliferation assay, briefly, 0.2×10^5 leukemic cells were incubated in 100 μl of RPMI red phenol-free medium (Thermo Fisher Scientific) in 96‐well plates. In initial screening assays, cells were incubated with 10 μM of each compound for 24, 48, and 72 h. To determine the concentration effect of the molecules, cells were treated with concentrations ranging from 100 nM to 100 μM for 48 h. Then, 0.1×10^5 HS27a cells and 0.035×10^5 MSCs were incubated in 100 µl of RPMI red phenol‐free medium in 96‐well plates for 48 h and then treated with concentrations ranging from 100 nM to 100 μM for 48 h.

Cells were incubated with 10 μ l of MTT working solution (5 g/l of methylthiazolyldiphenyl‐tetrazolium bromide from Sigma‐Aldrich) for 4 h. Cells were then lysed overnight at 37°C with 100 μl of 10% sodium dodecyl sulfate and 0.003% HCl. Optical density at 570 nm was measured using a spectrophotometer CLARIOstar® (BMG Labtech).

4.2.3 | FRET melting experiments

FRET melting experiments were run on a Stratagene Mx3005P real-time PCR equipment in 96-well plates on the DNA sequences reported in Table 1. Experiments were performed in 10 mM lithium cacodylate buffer (pH 7.2) and either 10 mM KCl and 90 mM LiCl (FK‐RAST, FBcl‐2T, F21T, and FdxT) or 1 mM KCl and 99 mM LiCl (Fc-MYCT) concentrations, depending on the T_m of the G4s alone (Table 5). The DNA concentration was 0.2 µM. The stabilization (ΔT_{m}) induced by compounds was calculated as the difference between the mid‐transition temperature of the nucleic acid alone and measured with the relevant ligand concentration, 2 µM. Data are presented as an average of three independent measurements, each conducted in duplicate conditions $(\lambda_{\text{exc}} = 492 \text{ nm}, \lambda_{\text{em}} = 516 \text{ nm}, T \text{ interval} = 25-95^{\circ}\text{C},$ ramp: 25°C for 5 min, then 1°C/min, measurements every 1°C, ×8 magnification of the fluorescence signal).

4.2.4 | Telomerase assays

Telomerase activity was assessed in two leukemic cell lines, MV4‐11 and HuT78. The TRAP assay (TRAPeze Telomerase Detection Kit; S7700; Millipore) was performed according to manufacturer's instructions, with some modifications. Briefly, $10⁶$ cells were resuspended in CHAPS lysis buffer and proteins were extracted. Protein extracts were used to extend a synthetic telomeric DNA by PCR amplification (1 cycle of 30°C for 30 min, telomere template extension with hotstart DNA polymerase activation, followed by a telomeric PCR amplification: 95°C for 3 min, 2 cycles of 95°C for 20 s, and 49°C for 20 s, and then 30 cycles of 95°C for 20 s and 60°C for 20 s with signal acquisition) in a Stratagene Mx3005P system (Agilent Technologies) using specific telomeric PCR quantitative primers.^[35] Each sample was measured in triplicate with a control DNA.

4.3 | Mass spectrometry

The affinity and stoichiometry of binding to G4‐forming oligonucleotide were determined by electrospray mass spectrometry in a negative ion mode on a Thermo Orbitrap Exactive mass spectrometer calibrated daily, and operated in negative mode, on a 350–2800 m/z scan range, using the 50,000 resolution setting, with the following tuning parameters: spray voltage: 3.1 kV, capillary temperature: 265°C, sheath gas: 50, aux gas: 0, heater temperature: 35°C, tube lens voltage: −185 V, capillary voltage: −50 V, skimmer voltage: −25 V. The syringe injection flow rate was 5 µl/ min. These parameters ensured a good signal intensity without disruption of the noncovalent complexes. Oligonucleotides were purchased lyophilized and RP-cartridge purified from Eurogentec, TMAA (1M) was purchased from Santa Cruz Biotechnology and potassium chloride (KCl > 99.99%) from Sigma‐Aldrich. The injected solutions were prepared by diluting the proper volume of the oligonucleotide stock solutions to reach 10 µM of DNA and 20 µM of ligand, in 100 mM TMAA and 1 mM KCl. The solutions were incubated for 16 h at 4°C before analysis.

The abundance of each species, M_iL_j , where M is a monomer oligonucleotide and L is a ligand, was determined from their peak area intensities, $I_{\text{Mil},i}$, from the 4⁻ and 5⁻ charge states:

$$
A_{M_i L_j} = \frac{I_{M_i L_j}}{\sum_{i=1,j=0}^{n,m} (i \times I_{M_i L_j})}.
$$

The molar concentrations were then calculated knowing the total amount of oligonucleotide $[M]_0$:

$$
[M_i L_j] = A_{M_i L_j} \times [M]_0 = \frac{I_{M_i L_j} \times [M]_0}{\sum_{i=1, j=0}^{n,m} \left(i \times I_{M_i L_j}\right)}
$$

The binding affinities for 1:1 and 2:1 L:M complex formation were determined from their dissociation constants:

$$
M + L \rightleftarrows ML; K_{d1} = \frac{[L]_{\text{free}} \times [M]_{\text{free}}}{[ML]},
$$

$$
ML + L \rightleftarrows ML; K_{d2} = \frac{[L]_{\text{free}} \times [ML]}{[ML_2]}.
$$

where the amount of free ligand is calculated from its total concentration $[L]_0$:

$$
\left[L\right]_{\text{free}} = \left[L_0\right] - \sum_{i=1,j=1}^{n,m} \left(j \times [M_i L_j]\right).
$$

The affinities were reported as $pK_d = -\log(K_d)$.

4.4 | Circular dichroism (CD)

CD experiments were performed with a JASCO J‐1500 spectropolarimeter using quartz cells of 2 mm path length. The scans were recorded at 22°C, from 220 to 350 nm with the following parameters: 1.0 nm data pitch, 2 nm bandwidth, 0.5 s response, 50 nm/min scanning

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speed; they are the result of three accumulations. Solutions were prepared as described in Section 4.3. The CD data were blank‐subtracted and then normalized to molar dichroic absorption (Δε, in cm⁻¹·M⁻¹) using the equation below, where $θ$ is the ellipticity in millidegrees, C is the oligonucleotide concentration in mol/l, and *l* is the path length in centimeters:

$$
\Delta \varepsilon = \frac{\theta}{32,980 \times C \times I}
$$

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CONFLICTS OF INTERESTS

The authors declare that there are no conflicts of interests.

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